## SCHEDULE 2

## METHODS OF ANALYSIS

## 10.

## DETERMINATION OF EXTRACTED PHOSPHORUS

## 7 EXPRESSION OF THE RESULTS

7．If the samples for analysis and dilutions shown in the Table are used the following formulae apply：

尼 $P$ in the lerriiiser $-(A-a) \times F^{\prime}$
络 $\mathrm{P}_{2} \mathrm{O}_{5}$ 加 the lertiliser $=(\mathrm{A}-\mathrm{a}) \times \mathrm{F}$
where：
$\mathrm{A}=$ weight in g of the quinoline phosphomolybdate
$\mathrm{a}=$ weight in g of the quinoline phosphomolybdate obtained in the blank test
F and $\mathrm{F}^{\prime}=$ factors given in the last two columns of the Table．
With samples for analysis and dilutions which differ from those of the Table the following formulae apply：－
\％ $\mathrm{P}_{2} \mathrm{O}_{5}$ in the lertilise：$-\frac{\left(\begin{array}{ll}\mathrm{A} & \text { a）} \times f \times \mathrm{D} \times 100 \\ \mathrm{M}\end{array} . . . . . .\right.}{}$

where：
$\mathrm{f}=$ conversion factor，quinoline phosphomolybdate into $\mathrm{P} 20 \mathrm{~s}=0.03074$
$\mathrm{f}^{\prime}=$ conversion factor，quinoline phosphomolybdate into $\mathrm{P}=0.013984$
$\mathrm{D}=$ dilution factor
$M=$ weighing of the sample analysed

| $\% \mathrm{P}_{2} \mathrm{O}_{5}$ in the fertiliser | \％P in the fertiliser | Sample <br> for <br> analysis <br> $g$ | Dilution to ml | Sample $m l$ | Dilution to ml | Sample Quinoline Quinoline <br> to be phosphomqdbbdahomolybdate <br> precipitatedonversionconversion  <br> $m l$ factor $\quad$ factor <br>  $(F)$ in <br>  $(F)$ in <br>  percentagepercentage <br>  $\mathrm{P}_{2} \mathrm{O}_{5}$ <br>  $P$ |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1－5 | 0．44－2．2 | 1 | 500 | － | － | 100 | 16.037 | 6.992 |
|  |  | 2.5 | 500 | － | － | 50 | 12.830 | 5.594 |
|  |  | 5 | 500 | － | － | 25 | 12.830 | 5.594 |
| 5－10 | 2．2－4．4 | 1 | 500 | － | － | 50 | 32.074 | 13.984 |
|  |  | 2.5 | 500 | － | － | 25 | 25.660 | 11.188 |

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