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(Acts whose publication is obligatory)

COMMISSION DIRECTIVE 95/31/EC

of 5 July 1995

laying down specific criteria of purity concerning sweeteners for use in foodstuffs

(Text with EEA relevance)

THE COMMISSION OF THE EUROPEAN COMMUNITIES,

Having regard to the Treaty establishing the European Community,

Having regard to Council Directive 89/107/EEC of 21 December 1988 on the approximation of the laws of the Member States concerning food additives authorized for use in foodstuffs intended for human consumption (¹), as amended by Directive 94/34/EC (²), and in particular Article 3 (3) (a) thereof,

After consultation of the Scientific Committee on Food,

Whereas it is necessary to establish purity criteria for all sweeteners mentioned in European Parliament and Council Directive 94/35/EC of 30 June 1994 on sweeteners for use in foodstuffs (³);

Whereas it is necessary to take into account the specifications and analytical techniques for sweeteners as set out in the *Codex Alimentarius* and the Joint FAO/WHO Expert Committee on Food Additives (Jecfa);

Whereas food additives, prepared by production methods or starting materials significantly different from those included in the evaluation of the Scientific Committee for Food, or different from those mentioned in this Directive, should be submitted for evaluation by the Scientific Committee for Food with a view to full evaluation with emphasis on the purity criteria; Whereas the measures provided for in this Directive are in line with the opinion of the Standing Committee on Foodstuffs,

HAS ADOPTED THIS DIRECTIVE:

Article 1

1. Purity criteria mentioned under Article 3 (3) (a) of Directive 89/107/EEC for sweeteners mentioned in Directive 94/35/EC are set out in the Annex.

2. The purity criteria for E 420 (i), E 420 (ii) and E 421 mentioned in the Annex to this Directive supersede the purity criteria for the said substances mentioned in the Annex to Council Directive 78/663/EEC (⁴).

Article 2

1. Member States shall bring into force the laws, regulations and administrative provisions necessary to comply with this Directive not later than 1 July 1996. They shall forthwith inform the Commission thereof.

When Member States adopt these provisions, these shall contain a reference to this Directive or shall be accompanied by such reference at the time of their official publication. The procedure for such reference shall be adopted by Member States.

⁽¹⁾ OJ No L 40, 11. 2. 1989, p. 27.

^{(&}lt;sup>2</sup>) OJ No L 237, 10. 9. 1994, p. 1.

^{(&}lt;sup>3</sup>) OJ No L 237, 10. 9. 1994, p. 3.

^{(&}lt;sup>4</sup>) OJ No L 223, 14. 8. 1978, p. 7.

2. Products put on the market or labelled before that date which do not comply with this Directive may, however, be marketed until stocks are exhausted.

Article 3

Done at Brussels, 5 July 1995.

Article 4

This Directive is addressed to the Member States.

This Directive shall enter into force on the 20th day following its publication in the Official Journal of the European Communities.

For the Commission Martin BANGEMANN Member of the Commission

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ANNEX

E 420 (i) - SORBITOL

Synonyms	D-glucitol, D-sorbitol
Definition	
Chemical name	D-glucitol
Einecs	200-061-5
E number	E 420 (i)
Chemical formula	$C_6H_{14}O_6$
Relative molecular mass	182,17
Assay	Content not less than 97% of total glycitols and not less than 91% of D-sorbitol on the dry weight basis.
	Glycitols are compounds with the structural formula CH_2OH -(CHOH) _n -CH ₂ OH, where 'n' is an integer
Description	White hygroscopic powder, crystalline powder, flakes or granules having a sweet taste
Identification	
A. Solubility	Very soluble in water, slightly soluble in ethanol
B. Melting range	88 to 102 °C
C. Sorbitol monobenzylidene derivative	To 5 g of the sample add 7 ml of methanol, 1 ml of benzaldehyde and 1 ml of hydrochloric acid. Mix and shake in a mechanical shaker until crystals appear. Filter with the aid of suction, dissolve the crystals in 20 ml of boiling water containing 1 g of sodium bicarbonate, filter while hot, cool the filtrate, filter with suction, wash with 5 ml of methanol-water mixture (1 in 2) and dry in air. The crystals so obtained melt between 173 and 179 °C
Purity	
Water content	Not more than 1 % (Karl Fischer method)
Sulphated ash	Not more than 0,1 % expressed on dry weight basis
Reducing sugars	Not more than 0,3% expressed as glucose on dry weight basis
Reducing sugars Total sugars	Not more than 0,3% expressed as glucose on dry weight basis Not more than 1% expressed as glucose on dry weight basis
Total sugars	Not more than 1% expressed as glucose on dry weight basis
Total sugars Chlorides	Not more than 1% expressed as glucose on dry weight basis Not more than 50 mg/kg expressed on dry weight basis

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Lead	Not more than 1 mg/kg expressed on dry weight basis	
Heavy metals	Not more than 10 mg/kg expressed as Pb on dry weight basis	
E 420 (ii) — SORBITOL SYRUP		
Synonyms	D-clucitol syrup	
Definition		-
Chemical name	Sorbitol syrup formed by hydrogenation of glucose syrup is composed of D-sorb D-mannitol and hydrogenated saccharides.	itol,
	The part of the product which is not D-sorbitol is composed mainly of hydrogeneral oligosaccharides formed by the hydrogenation of glucose syrup used as raw mater (in which case the syrup is non-crystallizing) or mannitol. Minor quantities of glyci where $n \leq 4$ may be present. Glycitols are compounds with the structural form CH_2OH -(CHOH) _n -CH ₂ OH, where 'n' is an integer	erial itols
Einecs	270-337-8	
E number	E 420 (ii)	
Assay	Content not less than 69% total solids and not less than 50% of D-sorbitol on anhydrous basis	the
Description	Clear colourless and sweet tasting aqueous solution	
Identification		
A. Solubility	Miscible with water, with glycerol, and with propane-1,2-diol	
B. Sorbitol monobenzylidene derivative	To 5 g of the sample add 7 ml of methanol, 1 ml of benzaldehyde and 1 m hydrochloric acid. Mix and shake in a mechanical shaker until crystals appear. F with the aid of suction, dissolve the crystals in 20 ml of boiling water containing of sodium bicarbonate, filter while hot. Cool the filtrate filter with suction, wash 5 ml of methanol-water mixture (1 in 2) and dry in air. The crystals so obtained between 173 and 179 °C	Filter ; 1 g with
Purity		
Water content	Not more than 31% (Karl Fischer method)	
Sulphated ash	Not more than 0,1 % expressed on dry weight basis	
Reducing sugars	Not more than 0,3% expressed as glucose on dry weight basis	
Chlorides	Not more than 50 mg/kg expressed on dry weight basis	
Sulphates	Not more than 100 mg/kg expressed on dry weight basis	
Nickel	Not more than 2 mg/kg expressed on dry weight basis	
Arsenic	Not more than 3 mg/kg expressed on dry weight basis	
Lead	Not more than 1 mg/kg expressed on dry weight basis	
Heavy metals	Not more than 10 mg/kg expressed as Pb on dry weight basis	

E 421 — MANNITOL

Synonyms	D-mannitol
Definition	
Chemical name	D-mannitol
Einecs	200-711-8
E number	E 421
Chemical formula	$C_6H_{14}O_6$
Relative molecular mass	182,2
Assay	Content not less than 96% D-mannitol on the dried basis
Description	Sweet tasting, white, odourless, crystalline powder
Identification	
A. Solubility	Soluble
Purity	
Loss on drying	Not more than 0,3% (105 °C, four hours)
pН	Between 5 and 8 Add 0,5 ml of a saturated solution of potassium chloride to 10 ml of a 10% w/v solution of the sample then measure the pH
Specific rotation	$(\alpha)_{\rm D}^{20}$ Specific rotation in a borated solution calculated with reference to the anhydrous substance is between + 23 and + 25°
Sulphated ash	Not more than 0,1 % expressed on dry weight basis
Reducing sugars	Not more than 0,3% expressed as glucose on dry weight basis
Total sugars	Not more than 1 % expressed as glucose on dry weight basis
Chlorides	Not more than 70 mg/kg expressed on dry weight basis
Sulphates	Not more than 100 mg/kg expressed on dry weight basis
Nickel	Not more than 2 mg/kg expressed on dry weight basis
Arsenic	Not more than 3 mg/kg expressed on dry weight basis
Lead	Not more than 1 mg/kg expressed on dry weight basis
Heavy metals	Not more than 10 mg/kg expressed as Pb on dry weight basis

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E 953 — ISOMALT			
Synonyms	Hydrogenated isomaltulose, hydrogenated palatinose		
Definition			
Chemical name	Isomalt is a mixture of: D-glucopyranosyl-1,6-D-glucitol and D-glucopyranosyl-1,1-D-mannitol dihydrate		
Einecs			
E number	E 953		
Chemical formula	D-glucopyranosyl-1,6-D-glucitol: D-glucopyranosyl-1,1-D-mannitol dihydrate:	$\begin{array}{c} C_{12}H_{24}O_{11}\\ C_{12}H_{24}O_{11}{\cdot}2H_2O\end{array}$	
Relative molecular mass	D-glucopyranosyl-1,6-D-glucitol: D-glucopyranosyl-1,1-D-mannitol dihydrate:	344,32 380,32	
Assay	Content lot less than 95% of the mixture of 1,6-D-glucitol and D-glucopyranosyl-1,1-D-mannitol dihydrai anhydrous basis	of D-glucopyranosyl- te determined on the	
Description	Odourless, white, sweet tasting, crystalline slightly hygroscopic	substance	
Identification			
A. Solubility	Slightly soluble in water, insoluble in ethanol		
B. Specific rotation	(α) ²⁰ _D : between + 90 and + 92° (4% w/v solution)		
C. Melting range	145 to 150 °C		
Tests			
Water content	Not more than 7% (Karl Fischer method)		
Sulphated ash	Not more than 0,05 expressed on dry weight basis		
Reducing sugars	Not more than 1,5 % expressed as glucose on dry weight basis	Not more than 1,5% expressed as glucose on dry weight basis	
Nickel	Not more than 2 mg/kg expressed on dry weight basis	Not more than 2 mg/kg expressed on dry weight basis	
Arsenic	Not more than 3 mg/kg expressed on dry weight basis		
Lead	Not more than 1 mg/kg expressed on dry weight basis		
Heavy metals	Not more than 10 mg/kg expressed on dry weight basis		

E 965 (i) — MALTITOL

Synonyms

Definition

Chemical name

D-maltitol, hydrogenated maltose

(a)-D-glucopyranosyl-1,4-D-glucitol

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Einecs	209-567-0	
E number	E 965 (i)	
Chemical formula	$C_{12}H_{24}O_{11}$	
Relative molecular mass	344,31	
Assay	Content not less than 98% D-mannitol $C_{12}H_{24}O_{11}$ on the and	nydrous basis
Description	Sweet tasting, white crystalline powder	. .
Identification	-	
A. Solubility	Very soluble in water, slightly soluble in ethanol	
B. Melting range	148 to 151 °C	
C. Specific rotation	$(\alpha)_{D}^{20} = + 105,5$ to + 105,5° (5% w/v solution)	
Purity		
Water content	Not more than 1% (Karl Fischer method)	
Sulphated ash	Not more than 0,1% expressed on dry weight basis	
Reducing sugars	Not more than 0,1% expressed as glucose on dry weight bas	is
Chlorides	Not more than 50 mg/kg expressed on dry weight basis	
Sulphates	Not more than 100 mg/kg expressed on dry weight basis	
Nickel	Not more than 2 mg/kg expressed on dry weight basis	
Arsenic	Not more than 3 mg/kg expressed on dry weight basis	
Lead	Not more than 1 mg/kg expressed on dry weight basis	
Heavy metals	Not more than 10 mg/kg expressed as Pb on dry weight basi	S

Hydrogenated high maltose-glucose syrup, hydrogenated glucose syrup

A mixture consisting of mainly maltitol with sorbitol and hydrogenated oligo- and polysaccharides. It is manufactured by the catalytic hydrogenation of high

maltose-content glucose syrup. The article of commerce is supplied both as a syrup and as a solid product

E 965 (ii) - MALTITOL SYRUP

Synonyms

Definition

Chemical name

Einecs

270-337-8

No L 178/8 EN	Official Journal of the European Communities 28. 7. 95
E number	E 965 (ii)
Assay	The following ranges apply on the anhydrous basis:
	Maltitol not less than 50%
	Sorbitol not more than 8 %
	Maltotriitol not more than 25 %
	Hydrogenated polysaccharides containing more than three glucose or glucitol units not not more than 30 %
Description	Sweet-tasting, colourless and odourless, clear viscous liquids or sweet-tasting white crystalline masses
Identification	
A. Solubility	Very soluble in water, slightly soluble in ethanol
B. Thin layer chromatography	Examine by the thin layer chromatography using a plate coated with a 0,25 mm layer of chromatographic silica gel
Purity	
Water content	Not more than 31 % (Karl Fischer method)
Sulphated ash	Not more than 0,1 % expressed on dry weight basis
Reducing sugars	Not more than 0,3 % expressed as glucose on dry weight basis
Chlorides	Not more than 50 mg/kg expressed on dry weight basis
Sulphates	Not more than 100 mg/kg expressed on dry weight basis
Nickel	Not more than 2 mg/kg expressed on dry weight basis
Arsenic	Not more than 3 mg/kg expressed on dry weight basis
Lead	Not more than 1 mg/kg expressed on dry weight basis
Heavy metals	Not more than 10 mg/kg expressed as Pb on dry weight basis
E 966 — LACTITOL	
Synonyms	Lactit, lactositol, lactobiosit
Definition	
Chemical name	4-O-β-D-galactopyranosyl-D-glucitol
Einecs	209-566-5
E number	E 966

Chemical formula

Relative molecular mass

Assay

Not less than 95% on the dry weight basis

 $C_{12}H_{24}O_{11}$

344,32

Description	Sweet-tasting crystalline powders or colourless solutions. Crystalline products occur in anhydrous, monohydrate and dihydrate forms
Identification	
A. Solubility	Very soluble in water
B. Specific rotation	$(\alpha)_D^{20}$ = + 13 to + 16° calculated on the anhydrous basis (10% w/v aqueous solution)
Purity	
Water content	Crystalline products; not more than 10,5 % (Karl Fischer method)
Other polyols	Not more than 2,5 % on the anhydrous basis
Reducing sugars	Not more than 0,2 % expressed as glucose on dry weight basis
Chlorides	Not more than 100 mg/kg expressed on dry weight basis
Sulphates	Not more than 200 mg/kg expressed on dry weight basis
Sulphated ash	Not more than 0,1 % expressed on dry weight basis
Nickel	Not more than 2 mg/kg expressed on dry weight basis
Arsenic	Not more than 3 mg/kg expressed on dry weight basis
Lead	Not more than 1 mg/kg expressed on dry weight basis
Heavy metals	Not more than 10 mg/kg expressed as Pb on dry weight basis
e 967 — Xylitol	
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Synonyms	Xylitol
Definition	
Chemical name	D-xylitol
Einecs	201-788-0
E number	E 967
Chemical formula	$C_5H_{12}O_5$
Relative molecular mass	152,15
Assay	Not less than 98,5% as xylitol on the anhydrous basis
Description	White, crystalline powder, practically odourless with a very sweet taste
Identification	
A. Solubility	Very soluble in water, sparingly soluble in ethanol
B. Melting range	92 to 96 °C
С. рН	5 to 7 (10% w/v aqueous solution)

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Purity	
Loss on drying	Not more than 0,5 %. Dry 0,5 g of sample in a vacuum over phosphorus at 60 °C for four hours
Sulphated ash	Not more than 0,1 % expressed on dry weight basis
Reducing sugars	Not more than 0,2 % expressed as glucose on dry weight basis
Other polyhydric alcohols	Not more than 1 % expressed on dry weight basis
Nickel	Not more than 2 mg/kg expressed on dry weight basis
Arsenic	Not more than 3 mg/kg expressed on dry weight basis
Lead	Not more than 1 mg/kg expressed on dry weight basis
Heavy metals	Not more than 10 mg/kg expressed as Pb on dry weight basis
Chlorides	Not more than 100 mg/kg expressed on dry weight basis
Sulphates	Not more than 200 mg/kg expressed on dry weight basis

E 950 — ACESULFAME K

Synonyms	Acesulfame potassium, acesulfam, potassium salt of 3,4-dihydro-6-methyl-1,2,3-oxathiazin-4-one-2,2-dioxide
Definition	
Chemical name	6-methyl-1,2,3-oxathiazin-4(3H)-one-2,2-dioxide potassium salt
Einecs	259-715-3
E number	E 950
Chemical formula	C ₄ H ₄ NO ₄ SK
Relative molecular mass	201,24
Assay	Not less than 99% of $C_4H_4NO_4SK$ on the anhydrous basis
Description	Odourless, white, crystalline powder having an intensively sweet taste. Aproximately 200 times as sweet as sucrose
Identification	
A. Solubility	Very soluble in water, very slightly soluble in ethanol
B. Ultra-violet absorption	Maximum 227 \pm 2 nm for a solution of 10 mg in 1 000 ml of water
Purity	
Loss on drying	Not more than 1% (105 °C, two hours)

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Arsenic	Not more than 3 r	ng/kg expressed on dry weight basis
Selenium	Not more than 30	mg/kg expressed on dry weight basis
Fluoride	Not more than 3 r	ng/kg expressed on dry weight basis
Lead	Not more than 1 r	ng/kg expressed on dry weight basis
Heavy metals	Not more than 10	mg/kg expressed as Pb on dry weight basis

E 951 — ASPARTAME

DefinitionIChemical nameN4-04 (Aspartyl-L-phenylalanine-1-methyl ester, 3-amino-N-tec-arbomethosy-phenethyl]-succinamic acid-N-methyl esterEnnecs245-261-3EnnemberE 951Chemical formulaCu _H I _N N ₂ OcRelative molecular mass294,31AssayNot less than 98% and not more than 102% of C ₁₊ H ₁₀ N ₂ Os on the anhydrous basisDescriptionWhite, odourless, crystalline powder having a sweet taste. Approximately 200 times as sweet as sucroseIdentificationPurityIsightly soluble in water and in ethanolPurityNot more than 0,2% expressed on dry weight basisJulp Addited ashNot more than 0,2% expressed on dry weight basispHBetween 4,5 and 6,0 (1 in 125 solution)PurityIn transmittanceglobalitier rotationNot more than 0,2%, equivalent to an absorbance of not more ethan 0,2%, equivalent to an absorbance of not more ethan 0,2%, equivalent to an absorbance of not more ethan 0,2%, equivalent to an absorbance of not more ethan 0,2%, equivalent to an absorbance of not more ethan 0,2%, equivalent to an absorbance of not more ethan 0,2%, equivalent to an absorbance of not more ethan 0,2%, equivalent to an absorbance of not more ethan 0,2%, equivalent to an absorbance of not more ethan 0,2%, equivalent to an absorbance of not more ethan 0,2%, equivalent to an absorbance of not more ethan 0,2%, equivalent to an absorbance of not more ethan 0,2%, equivalent to an absorbance of not more ethan 0,2%, equivalent to an absorbance of not more ethan 0,2%, equivalent to an absorbance of not more ethan 0,2%, equivalent to an absorbance of not more ethan 0,2%, equivalent to an absorbance of not more ethan 0,2%, equivalent to an absorbance of not mor	Synonyms	Aspartyl phenylalanine methyl ester
J-amino-N-(a-carbomethoxy-phenethyl)-succinamic acid-N-methyl esterEinecs245-261-3E numberE 951Chemical formulaC ₁₄ H ₁₀ N ₂ O ₅ Relative molecular mass294,31AssayNot less than 98% and not more than 102% of C ₁₄ H ₁₀ N ₂ O ₅ on the anhydrous basisDescriptionWhite, odourless, crystalline powder having a sweet taste. Approximately 200 times as sweet as sucroseIdentificationImage: Construction of the sample solutionPurityImage: Construction of the sample solution in 2N hydrochloric acid, determined in a 1-cm cell at 430 nm with a suitable spectrophorometer, using 2N hydrochloric acid as a reference, is not less than 0,5% convertient of an absorbance of not more than approximately 0,022Specific rotation(a) ¹⁰ / ₁₀ + 14,5 to +16,5° Determine in a 4 in 100/15 N formic acid solution within 30 minutes after preparation of the sample solution	Definition	
E numberE 951Chemical formula $C_{14}H_{18}N_2O_7$ Relative molecular mass294,31AssayNot less than 98% and not more than 102% of $C_{14}H_{18}N_2O_3$ on the anhydrous basisDescriptionWhite, odourless, crystalline powder having a sweet taste. Approximately 200 times as sweet as sucroseIdentificationImage: Comparison of the substraint of the substra	Chemical name	
Chemical formula $C_{14}H_{18}N_3O_5$ Relative molecular mass294,31AssayNot less than 98% and not more than 102% of $C_{12}H_{18}N_3O_5$ on the anhydrous basisDescriptionWhite, odourless, crystalline powder having a sweet taste. Approximately 200 times as sweet as sucroseIdentificationImage: SolubilitySolubilitySlightly soluble in water and in ethanolPurityImage: SolubilityLoss on dryingNot more than 0,2% expressed on dry weight basispHBetween 4,5 and 6,0 (1 in 125 solution)TransmittanceThe transmittance of a 1% solution in 2N hydrochloric acid, determined in a 1-cm cell at 430 nm with a suitable spectrophotometer, using 2N hydrochloric acid as a reference, is not less than 0,95, equivalent to an absorbance of not more than approximately 0,022Specific rotation $(\alpha)_{13}^{\alpha_1}: 14.5$ to $+16.5^{\alpha}$ Determine in a 4 in 100/15 N formic acid solution within 30 minutes after preparation of the sample solution	Einecs	245-261-3
Relative molecular mass294,31AssayNot less than 98% and not more than 102% of $C_{14}H_{18}N_2O_5$ on the anhydrous basisDescriptionWhite, odourless, crystalline powder having a sweet taste. Approximately 200 times as sweet as sucroseIdentificationIdentificationSolubilitySlightly soluble in water and in ethanolPurityIdent than 4,5% (105 °C, four hours)Sulphated ashNot more than 4,5% (105 °C, four hours)Sulphated ashNot more than 0,2% expressed on dry weight basispHBetween 4,5 and 6,0 (1 in 125 solution)TransmittanceThe transmittance of a 1% solution in 2N hydrochloric acid, determined in a 1-cm cell at 430 nm with a suitable spectrophotometer, using 2N hydrochloric acid as a reference, is not less than 0,95, equivalent to an absorbance of not more than approximately 0,022Specific rotation $(\alpha)_{13}^{23}: +14,5$ to $+16,5^{\circ}$ Determine in a 4 in 100/15 N formic acid solution within 30 minutes after preparation of the sample solution	E number	E 951
AssayNot less than 98% and not more than 102% of $C_{14}H_{18}N_2O_5$ on the anhydrous basisDescriptionWhite, odourless, crystalline powder having a sweet taste. Approximately 200 times as sweet as sucroseIdentificationSlightly soluble in water and in ethanolSolubilitySlightly soluble in water and in ethanolPurityLoss on dryingNot more than 4,5% (105 °C, four hours)Sulphated ashNot more than 0,2% expressed on dry weight basispHBetween 4,5 and 6,0 (1 in 125 solution)TransmittanceThe transmittance of a 1% solution in 2N hydrochloric acid, determined in a 1-cm arference, is not less than 0,95, equivalent to an absorbance of not more than approximately 0,022Specific rotation($\alpha_1^{30}_{12}$: 14,5 to +16,5° Determine in a 4 in 100/15 N formic acid solution within 30 minutes after preparation of the sample solution	Chemical formula	$C_{14}H_{18}N_{2}O_{5}$
basisDescriptionWhite, odourless, crystalline powder having a sweet taste. Approximately 200 times as sweet as sucroseIdentificationSlightly soluble in water and in ethanolSolubilitySlightly soluble in water and in ethanolPurityNot more than 4,5 % (105 °C, four hours)Sulphated ashNot more than 0,2 % expressed on dry weight basispHBetween 4,5 and 6,0 (1 in 125 solution)TransmittanceThe transmittance of a 1 % solution in 2N hydrochloric acid, determined in a 1-cm cell at 430 nm with a suitable spectrophotometer, using 2N hydrochloric acid as a reference, is not less than 0,95, equivalent to an absorbance of not more than approximately 0,022Specific rotation $(\alpha)_{11}^{\alpha_1} : 14,5$ to $+16,5^{\circ}$ Determine in a 4 in 100/15 N formic acid solution within 30 minutes after preparation of the sample solution	Relative molecular mass	294,31
IdentificationSolubilitySolubilityPurityLoss on dryingNot more than 4,5% (105 °C, four hours)Sulphated ashPHBetween 4,5 and 6,0 (1 in 125 solution)TransmittanceTransmittanceSpecific rotation $(\alpha)_{11}^{30}: +14,5$ to $+16,5^{\circ}$ Determine in a 4 in 100/15 N formic acid solution within 30 minutes after preparation of the sample solution	Assay	
SolubilitySlightly soluble in water and in ethanolPurityI.oss on dryingNot more than 4,5 % (105 °C, four hours)Sulphated ashNot more than 0,2 % expressed on dry weight basispHBetween 4,5 and 6,0 (1 in 125 solution)TransmittanceThe transmittance of a 1 % solution in 2N hydrochloric acid, determined in a 1-cm cell at 430 nm with a suitable spectrophotometer, using 2N hydrochloric acid as a reference, is not less than 0,95, equivalent to an absorbance of not more than approximately 0,022Specific rotation(a) 30/15 t 14,5 to +16,5° Determine in a 4 in 100/15 N formic acid solution within 30 minutes after preparation of the sample solution	Description	
Purity Loss on drying Not more than 4,5 % (105 °C, four hours) Sulphated ash Not more than 0,2 % expressed on dry weight basis pH Between 4,5 and 6,0 (1 in 125 solution) Transmittance The transmittance of a 1 % solution in 2N hydrochloric acid, determined in a 1-cm cell at 430 nm with a suitable spectrophotometer, using 2N hydrochloric acid as a reference, is not less than 0,95, equivalent to an absorbance of not more than approximately 0,022 Specific rotation (a) ²⁰ / ₂₀ : +14,5 to +16,5° Determine in a 4 in 100/15 N formic acid solution within 30 minutes after preparation of the sample solution	Identification	
Loss on dryingNot more than 4,5 % (105 °C, four hours)Sulphated ashNot more than 0,2 % expressed on dry weight basis pH Between 4,5 and 6,0 (1 in 125 solution)TransmittanceThe transmittance of a 1 % solution in 2N hydrochloric acid, determined in a 1-cm cell at 430 nm with a suitable spectrophotometer, using 2N hydrochloric acid as a reference, is not less than 0,95, equivalent to an absorbance of not more than approximately 0,022Specific rotation $(\alpha)_{10}^{20}$: +14,5 to +16,5° Determine in a 4 in 100/15 N formic acid solution within 30 minutes after preparation of the sample solution	Solubility	Slightly soluble in water and in ethanol
Sulphated ashNot more than $0,2$ % expressed on dry weight basispHBetween 4,5 and 6,0 (1 in 125 solution)TransmittanceThe transmittance of a 1% solution in 2N hydrochloric acid, determined in a 1-cm cell at 430 nm with a suitable spectrophotometer, using 2N hydrochloric acid as a reference, is not less than 0,95, equivalent to an absorbance of not more than approximately 0,022Specific rotation $(\alpha)_D^{20}: +14,5$ to $+16,5^\circ$ Determine in a 4 in 100/15 N formic acid solution within 30 minutes after preparation of the sample solution	Purity	
pHBetween 4,5 and 6,0 (1 in 125 solution)TransmittanceThe transmittance of a 1% solution in 2N hydrochloric acid, determined in a 1-cm cell at 430 nm with a suitable spectrophotometer, using 2N hydrochloric acid as a reference, is not less than 0,95, equivalent to an absorbance of not more than approximately 0,022Specific rotation $(\alpha)_D^{20}: +14,5$ to $+16,5^\circ$ Determine in a 4 in 100/15 N formic acid solution within 30 minutes after preparation of the sample solution	Loss on drying	Not more than 4,5 % (105 °C, four hours)
TransmittanceThe transmittance of a 1% solution in 2N hydrochloric acid, determined in a 1-cm cell at 430 nm with a suitable spectrophotometer, using 2N hydrochloric acid as a reference, is not less than 0,95, equivalent to an absorbance of not more than approximately 0,022Specific rotation $(\alpha)_D^{20}: +14,5$ to $+16,5^\circ$ Determine in a 4 in 100/15 N formic acid solution within 30 minutes after preparation of the sample solution	Sulphated ash	Not more than 0,2% expressed on dry weight basis
cell at 430 nm with a suitable spectrophotometer, using 2N hydrochloric acid as a reference, is not less than 0,95, equivalent to an absorbance of not more than approximately 0,022 Specific rotation $(\alpha)_{D}^{20}: +14,5 \text{ to } +16,5^{\circ}$ Determine in a 4 in 100/15 N formic acid solution within 30 minutes after preparation of the sample solution	pН	Between 4,5 and 6,0 (1 in 125 solution)
Determine in a 4 in 100/15 N formic acid solution within 30 minutes after preparation of the sample solution	Transmittance	cell at 430 nm with a suitable spectrophotometer, using 2N hydrochloric acid as a reference, is not less than 0.95 , equivalent to an absorbance of not more than
Arsenic Not more than 3 mg/kg expressed on dry weight basis	Specific rotation	Determine in a 4 in 100/15 N formic acid solution within 30 minutes after
	Arsenic	Not more than 3 mg/kg expressed on dry weight basis
Lead Not more than 1 mg/kg expressed on dry weight basis	Lead	Not more than 1 mg/kg expressed on dry weight basis

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Heavy metals

Not more than 10 mg/kg expressed as Pb on dry weight basis

5-Benzyl-3,6-dioxo-2-piperazineacetic acid

Not more than 1,5 % expressed on dry weight basis

E 952 - CYCLAMIC ACID AND ITS Na AND Ca SALTS

(I) CYCLAMIC ACID **Synonyms** Cyclohexylsulphamic acid, cyclamate Definition Chemical name Cyclohexanesulphamic acid, cyclohexylaminosulphonic acid 202-898-1 **Einecs** E number E 952 Chemical formula $C_6H_{13}NO_3S$ Relative molecular mass 179,24 Cyclohexylsulphamic acid contains not less than 98% and not more than the Assay equivalent of 102% of C₆H₁₃NO₃S, calculated on the anhydrous basis Description A practically colourless, white crystalline powder with a sweet-sour taste. Approximately 40 times as sweet as sucrose Identification A. Solubility Soluble in water and in ethanol Acidify a 2% solution with hydrochloric acid, add 1 ml of an approximately molar B. Precipitation test solution of barium chloride in water and filter if any haze or precipitate forms. To the clear solution add 1 ml of a 10% solution of sodium nitrite. A white precipitate forms. Purity Loss on drying Not more than 1 % (105 °C, one hour) Selenium Not more than 30 mg/kg expressed as selenium on dry weight basis Lead Not more than 1 mg/kg expressed on dry weight basis Heavy metals Not more than 10 mg/kg expressed as Pb on dry weight basis Arsenic Not more than 3 mg/kg expressed on dry weight basis Cyclohexylamine Not more than 10 mg/kg expressed on dry weight basis Dicyclohexylamine Not more than 1 mg/kg expressed on dry weight basis

Not more than 1 mg/kg expressed on dry weight basis

Aniline

(II) SODIUM CYCLAMATE	
Synonyms	Cyclamate, sodium salt of cyclamic acid
Definition	
Chemical name	Sodium cyclohexanesulphamate, sodium cyclohexylsulphamate
Einecs	205-348-9
E number	E 952
Chemical formula	C ₆ H ₁₂ NNaO ₃ S and the dihydrate form C ₆ H ₁₂ NNaO ₃ S·2H ₂ O
Relative molecular mass	201,22 calculated on the anhydrous form 237,22 calculated on the hydrated form
Assay	Not les sthan 98% and not more than 102% on the dried basis Dihydrate form: not less than 84% on the dried basis
Description	White, odourless crystals or crystalline powder. Approximately 30 times as sweet as sucrose
Identification	
Solubility	Soluble in water, practically insoluble in ethanol
Purity	
Loss on drying	Not more than 1 % (105 °C, one hour) Not more than 15,2 % (105 °C, two hours) for the dihydrate form
Selenium	Not more than 30 mg/kg expressed as selenium on dry weight basis
Arsenic	Not more than 3 mg/kg expressed on dry weight basis
Lead	Not more than 1 mg/kg expressed on dry weight basis
Heavy metals	Not more than 10 mg/kg expressed as Pb on dry weight basis
Cyclohexylamine	Not more than 10 mg/kg expressed on dry weight basis
Dicyclohexylamine	Not more than 1 mg/kg expressed on dry weight basis
Aniline	Not more than 1 mg/kg expressed on dry weight basis
(III) CALCIUM CYCLAMATE	
Synonyms	Cyclamate, calcium salt of cyclamic acid
Definition	
Chemical name	Calcium cyclohexanesulphamate, calcium cyclohexylsulphamate
Einecs	205-349-4
E number	E 952
Chemical formula	$C_{12}H_{24}CaN_2O_6S_2\cdot 2H_2O$

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Relative molecular mass	432,57	
Assay	Not less than 98% and not more than 10% on the dried basis	S
Description	White, colourless crystals or crystaline powder. Approximately sucrose	30 times as sweet as
Identification		
Solubility	Soluble in water, sparingly soluble in ethanol	
Purity		
Loss on drying	Not more than 1% (105 °C, one hour) Not more than 8,5% (140 °C, four hours) for the dihydrate fo	orm
Selenium	Not more than 30 mg/kg expressed as selenium on dry weight	basis
Arsenic	Not more than 3 mg/kg expressed on dry weight basis	
Lead	Not more than 1 mg/kg expressed on dry weight basis	
Heavy metals	Not more than 10 mg/kg expressed as Pb on dry weight basis	
Cyclohexylamine	Not more than 10 mg/kg expressed on dry weight basis	
Dicycloh exylamin e	Not more than 1 mg/kg expressed on dry weight basis	
Aniline	Not more than 1 mg/kg expressed on dry weight basis	

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E 954 - SACCHARIN AND ITS Na. K AND Ca SALTS

(I) SACCHARIN

Definition		
Chemical name	3-oxo-2,3dihydrobenzo(d)isothiazol-1,1-dioxide	
Einecs	201-321-0	
E number	E 954	
Chemical formula	$C_7H_5NO_3S$	
Relative molecular mass	183,18	
Assay	Not less than 99% and not more than 101,0% of $C_7H_5NO_3S$ on the anhydrous basis	
Description	White crystals or a white crystalline powder, odourless or with a faint, aromatic odour having a sweet taste even in very dilute solutions. Approximately between 300 and 500 times as sweet as sucrose	
Identification		
Solubility	Slightly soluble in water, soluble in basic solutions, sparingly soluble in ethanol	

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Purity	
Loss on drying	Not more than 1 % (105 °C, two hours)
Melting range	226 to 230 °C
Arsenic	Not more than 3 mg/kg expressed on dry weight basis
Selenium	Not more than 30 mg/kg expressed on dry weight basis
Lead	Not more than 1 mg/kg expressed on dry weight basis
Heavy metals	Not more than 10 mg/kg expressed as Pb on dry weight basis
Sulphated ash	Not more than 0,2% expressed on dry weight basis
Benzoic and salicylic acid	To 10 ml of a 1 in 20 solution, previously acidified with five drops of acetic acid, add three drops of an approximately molar solution of ferric chloride in water. No precipitate or violet colour appears
o-Toluenesulphonamide	Not more than 10 mg/kg expressed on dry weight basis
p-Toluenesulphonamide	Not more than 10 mg/kg expressed on dry weight basis
Benzoic acid p-sulfonamide	Not more than 25 mg/kg expressed on dry weight basis
Readily carbonizable substances	Absent
(II) SODIUM SACCHARIN	
Synonyms	Saccharin, sodium salt of saccharin
Definition	
Chemical name	Sodium o-benzosulphimide, sodium salt of 2,3-dihydro-3-oxobenzisosulfonazole, oxobenzisosulfonazole, 1,2-benzisothiazolin-3-one-1, 1-dioxide sodium salt dihydrate
Einecs	204-886-1
E number	E 954
Chemical formula	$C_7H_4NNaO_3S\cdot 2H_2O$
Relative molecular mass	241,19
Assay	Not less than 99% and not more than 101% of $C_7H_4NNaO_3S$ on the anhydrous basis
Description	White crystals or a white crystalline efflorescent powder, odourless or with a faint, odour, hvaing an intensely sweet taste, even in very dilute solutions. Approximately between 300 and 500 times as sweet as sucrose in dilute solutions
Identification	
Solubility	Freely soluble in water, sparingly soluble in ethanol
Purity	
Loss on drying	Not more than 15% (120 °C, four hours)

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Arsenic	Not more than 3 mg/kg expressed on dry weight basis	
Selenium	Not more than 30 mg/kg expressed on dry weight basis	
Lead	Not more than 1 mg/kg expressed on dry weight basis	
Heavy metals	Not more than 10 mg/kg expressed as Pb on dry weight basis	
Benzoic and salicylic acid	To 10 ml of a 1 in 20 solution, previously acidified with five drop three drops of an approximately molar solution of ferric chl precipitate or violet colour appears	
o-Toluenesulphonamide	Not more than 10 mg/kg expressed on dry weight basis	
p-Toluenesulphonamide	Not more than 10 mg/kg expressed on dry weight basis	
Benzoic acid p-sulfonamide	Not more than 25 mg/kg expressed on dry weight basis	
Readily carbonizable substances	Absent	
III) CALCIUM SACCHARIN		
Synonyms	Saccharin, calcium salt of saccharin	
Definition		
Chemical name	Calcium o-benzosulphimide, calcium salt of 2,3-dihydro-3-oxobenzisosulphonazole, 1,2-benzisothiazolin-3-one-1,1-dioxide calcium salt hydrate (2:7)	
Einecs	229-349-0	
E number	E 954	
Chemical formula	$C_{14}H_8CaN_2O_6S_2\cdot 3^1/_2H_2O$	
Relative molecular mass	467,48	
Assay	Not less than 95 % of $C_{14}H_8\text{Ca}\text{N}_2\text{O}_6\text{S}_2$ on the anhydrous basis	
Description	White crystals or a white crystalline powder, odourless or with a an intensely sweet taste, even in very dilute solutions. Approximat 500 times as sweet as sucrose in dilute solutions	faint odour, havin ely between 300 an
Identification		
Solubility	Freely soluble in water, soluble in ethanol	-
Purity		
Loss on drying	Not more than 13,5% (120 °C, four hours)	
Arsenic	Not more than 3 mg/kg expressed on dry weight basis	
Selenium	Not more than 30 mg/kg expressed on dry weight basis	
Lead	Not more than 1 mg/kg expressed on dry weight basis	
Heavy metals	Not more than 10 mg/kg expressed as Pb on dry weight basis	

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	Benzoic and salicylic acid	To 10 ml of a 1 in 20 solution, previously acidified with five drops of acetic acid, ad three drops of an approximately molar solution of ferric chloride in water. N precipitate or violet colour appears
	o-Toluenesulphonamide	Not more than 10 mg/kg expressed on dry weight basis
	p-Toluenesulphonamide	Not more than 10 mg/kg expressed on dry weight basis
	Benzoic acid p-sulfonamide	Not more than 25 mg/kg expressed on dry weight basis
	Readily carbonizable substances	Absent
(IV)	POTASSIUM SACCHARIN	
	Synonyms	Saccharin, potassium salt of saccharin
	Definition	
	Chemical name	Potassium o-benzosulphimide, potassium salt of 2,3-dihydro-3-oxobenzisosulphonazole, potassium salt of 1,2-benzisothiazolin-3-one-1,1-dioxide monohydrate
	Einecs	
	E number	E 954
	Chemical formula	C ₇ H ₄ KNO ₃ S·H ₂ O
	Relative molceular mass	239,77
	Assay	Not less than 99% and not more than 101% of $C_7H_4KNO_3S$ on the anhydron basis
	Description	White crystals or a white crystalline powder, odourless or with a faint odour, havin an intensely sweet taste, even in very dilute solutions. Approximately between 300 ar 500 times as sweet as sucrose
	Identification	
	Solubility	Freely soluble in water, sparingly soluble in ethanol
	Purity	
	Loss on drying	Not more than 8% (120 °C, four hours)
	Arsenic	Not more than 3 mg/kg expressed on dry weight basis
	Selenium	Not more than 30 mg/kg expressed on dry weight basis
	Lead	Not more than 1 mg/kg expressed on dry weight basis
	Heavy metals	Not more than 10 mg/kg expressed as Pb on dry weight basis
	Benzoic and salicylic acid	To 10 ml of a 1 in 20 solution, previously acidified with five drops of acetic acid, ac three drops of an approximately molar solution of ferric chloride in water. N precipitate or violet colour appears
	o-Toluenesulphonamide	Not more than 10 mg/kg expressed on dry weight basis
	p-Toluenesulphonamide	Not more than 10 mg/kg expressed on dry weight basis

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Benzoic acid p-sulfonamide	Not more than 25 mg/kg expressed on dry weight basis
Readily carbonizable substances	Absent
E 957 — THAUMATIN	
Synonyms	
Definition	
Chemical name	Thaumatin is obtained by aqueous extraction (pH 2,5 to 4) of the arils of the fruit of the natural strain of <i>Thaumatococcus daniellii</i> (Benth) and consists essentially of the proteins thaumatin I and thaumatin II together with minor amounts of plant constituents derived from the source material
Einecs	258-822-2
E number	E 957
Chemical formula	Polypeptide of 207 aminoacids
Relative molecular mass	Thaumatin I 22209 Thaumatin II 22293
Assay	Not less than 16% nitrogen on the dried basis equivalent to not less than 94% proteins (N \times 5,8)
Description	Odourless, cream-coloured powder with an intensely sweet taste. Approximately 2 000 to 3 000 times as sweet as sucrose
Identification	
Solubility	Very soluble in water, insoluble in acetone
Purity	
Loss on drying	Not more than 9% (105 °C to constant weight)
Carbohydrates	Not more than 3% expressed on dry weight basis
Sulphated ash	Not more than 2 % expressed on dry weight basis
Aluminium	Not more than 100 mg/kg expressed on dry weight basis
Arsenic	Not more than 3 mg/kg expressed on dry weight basis
Lead	3 mg/kg expressed on dry weight basis
Microbiological criteria	Total aerobic microbial count: Max 1 000/g E. Coli: absent in 1 g

E 959 — NEOHESPERIDINE DIHYDROCHALCONE

Synonyms	Neohesperidin dihydrochalcone, NHDC, he neohesperidoside, neohesperidin DC	speretin	dihydrochalcone-4'-β-
Definition			
Chemical name	2-O-α-L-rhamnopyranosyl-4′-β-D-glucopyranosyl obtained by catalytic hydrogenation of neohesperidi	hesperetin 1	dihydrochalcone;

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Einecs	243-978-6	
E number	E 959	
Chemical formula	$C_{28}H_{36}O_{15}$	
Relative molecular mass	612,6	
Assay	Content not less than 96% on the dried basis	
Description	Off white, odourless, crystalline powder having a characteristic, intensive sweet taste Approximately between 1 000 and 1 800 times as sweet as sucrose	
Identification		
A. Solubility	Freely soluble in hot water, very slightly soluble in cold w ether and benzene	rater, practically insoluble in
B. Ultraviolet absorption maximum	282 to 283 nm for a solution of 2 mg in 100 ml methan	nol .
C. Neu's test	Dissolve about 10 mg of neohesperidine DC in 1 ml me 2-aminoethyl diphenyl borate methanolic solution. A produced	
Purity		
Loss on drying	Not more than 11% (105 °C, three huors)	
Sulphated ash	Not more than 0,2% expressed on dry weight basis	
Arsenic	Not more than 3 mg/kg expressed on dry weight basis	
Lead	Not more than 2 mg/kg expressed on dry weight basis	
Heavy metals	Not more than 10 mg/kg expressed as Pb on dry weight	basis

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