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COMMISSION DIRECTIVE 95/31/EC

of 5 July 1995

laying down specific criteria of purity concerning sweeteners for use in foodstuffs

(Text with EEA relevance)

(OJ L 178, 28.7.1995, p. 1)

Amended by:

<u>▶</u>B

Official Journal

		No	page	date
<u>M1</u>	Commission Directive 98/66/EC of 4 September 1998	L 257	35	19.9.1998
► <u>M2</u>	Commission Directive 2000/51/EC of 26 July 2000	L 198	41	4.8.2000
► <u>M3</u>	Commission Directive 2001/52/EC of 3 July 2001	L 190	18	12.7.2001
► <u>M4</u>	Commission Directive 2004/46/EC of 16 April 2004	L 114	15	21.4.2004
►M5	Commission Directive 2006/128/EC of 8 December 2006	L 346	6	9.12.2006

COMMISSION DIRECTIVE 95/31/EC

of 5 July 1995

laying down specific criteria of purity concerning sweeteners for use in foodstuffs

(Text with EEA relevance)

THE COMMISSION OF THE EUROPEAN COMMUNITIES,

Having regard to the Treaty establishing the European Community,

Having regard to Council Directive 89/107/EEC of 21 December 1988 on the approximation of the laws of the Member States concerning food additives authorized for use in foodstuffs intended for human consumption (1), as amended by Directive 94/34/EC (2), and in particular Article 3 (3) (a) thereof,

After consultation of the Scientific Committee on Food,

Whereas it is necessary to establish purity criteria for all sweeteners mentioned in European Parliament and Council Directive 94/35/EC of 30 June 1994 on sweeteners for use in foodstuffs (3);

Whereas it is necessary to take into account the specifications and analytical techniques for sweeteners as set out in the Codex Alimentarius and the Joint FAO/WHO Expert Committee on Food Additives (Jecfa):

Whereas food additives, prepared by production methods or starting materials significantly different from those included in the evaluation of the Scientific Committee for Food, or different from those mentioned in this Directive, should be submitted for evaluation by the Scientific Committee for Food with a view to full evaluation with emphasis on the purity criteria;

Whereas the measures provided for in this Directive are in line with the opinion of the Standing Committee on Foodstuffs,

HAS ADOPTED THIS DIRECTIVE:

Article 1

- Purity criteria mentioned under Article 3 (3) (a) of Directive 89/107/EEC for sweeteners mentioned in Directive 94/35/EC are set out in the Annex.
- The purity criteria for E 420 (i), E 420 (ii) and E 421 mentioned in the Annex to this Directive supersede the purity criteria for the said substances mentioned in the Annex to Council Directive 78/663/EEC $(^{4}).$

Article 2

Member States shall bring into force the laws, regulations and administrative provisions necessary to comply with this Directive not later than 1 July 1996. They shall forthwith inform the Commission thereof.

⁽¹⁾ OJ No L 40, 11. 2. 1989, p. 27. (2) OJ No L 237, 10. 9. 1994, p. 1. (3) OJ No L 237, 10. 9. 1994, p. 3.

⁽⁴⁾ OJ No L 223, 14. 8. 1978, p. 7.

▼<u>B</u>

When Member States adopt these provisions, these shall contain a reference to this Directive or shall be accompanied by such reference at the time of their official publication. The procedure for such reference shall be adopted by Member States.

2. Products put on the market or labelled before that date which do not comply with this Directive may, however, be marketed until stocks are exhausted.

Article 3

This Directive shall enter into force on the 20th day following its publication in the *Official Journal of the European Communities*.

Article 4

This Directive is addressed to the Member States.

ANNEX

E 420 (i) — SORBITOL

Synonyms D-glucitol, D-sorbitol

Definition

Chemical name D-glucitol Einecs 200-061-5 E number E 420 (i) Chemical formula $C_6H_{14}O_6$ 182,17 Relative molecular mass

Content not less than 97% of total glycitols and not less than Assay

91% of D-sorbitol on the dry weight basis.

Glycitols are compounds with the structural formula CH2OH-

(CHOH)_n-CH₂OH, where 'n' is an integer

Description White hygroscopic powder, crystalline powder, flakes or

granules having a sweet taste

Identification

A. Solubility Very soluble in water, slightly soluble in ethanol

B. Melting range 88 to 102°C

C. Sorbitol monobenzylidene derivative

To 5 g of the sample add 7 ml of methanol, 1 ml of benzaldehyde and 1 ml of hydrochloric acid. Mix and shake in a mechanical shaker until crystals appear. Filter with the aid of suction, dissolve the crystals in 20 ml of boiling water containing 1 g of sodium bicarbonate, filter while hot, cool the filtrate, filter with suction, wash with 5 ml of methanolwater mixture (1 in 2) and dry in air. The crystals so obtained

melt between 173 and 179°C

Purity

Water content Not more than 1% (Karl Fischer method)

Sulphated ash Not more than 0,1% expressed on dry weight basis

Not more than 0,3% expressed as glucose on dry weight basis Reducing sugars Total sugars Not more than 1% expressed as glucose on dry weight basis Chlorides Not more than 50 mg/kg expressed on dry weight basis Sulphates Not more than 100 mg/kg expressed on dry weight basis Nickel Not more than 2 mg/kg expressed on dry weight basis Arsenic Not more than 3 mg/kg expressed on dry weight basis Lead Not more than 1 mg/kg expressed on dry weight basis

Heavy metals Not more than 10 mg/kg expressed as Pb on dry weight basis

E 420 (ii) — SORBITOL SYRUP

Synonyms D-clucitol syrup

Definition

Chemical name

Sorbitol syrup formed by hydrogenation of glucose syrup is composed of D-sorbitol, D-mannitol and hydrogenated saccharides.

The part of the product which is not D-sorbitol is composed mainly of hydrogenated oligosaccharides formed by the hydrogenation of glucose syrup used as raw material (in which case the syrup is non-crystallizing) or mannitol. Minor quantities of glycitols where $n \le 4$ may be present. Glycitols are compounds with the structural formula CH_2OH - $(CHOH)_n$ - CH_2OH , where 'n' is an integer

Einecs 270-337-8 E number E 420 (ii)

Assay Content not less than 69% total solids and not less than 50% of

D-sorbitol on the anhydrous basis

Description Clear colourless and sweet tasting aqueous solution

▼B

Identification

A. Solubility

B. Sorbitol monobenzylidene derivative

Miscible with water, with glycerol, and with propane-1,2-diol

To 5 g of the sample add 7 ml of methanol, 1 ml of benzaldehyde and 1 ml of hydrochloric acid. Mix and shake in a mechanical shaker until crystals appear. Filter with the aid of suction, dissolve the crystals in 20 ml of boiling water containing 1 g of sodium bicarbonate, filter while hot. Cool the filtrate filter with suction, wash with 5 ml of methanolwater mixture (1 in 2) and dry in air. The crystals so obtained melt between 173 and 179°C

Purity

Not more than 31% (Karl Fischer method) Water content

Sulphated ash Not more than 0,1% expressed on dry weight basis

Reducing sugars Not more than 0,3% expressed as glucose on dry weight basis Chlorides Not more than 50 mg/kg expressed on dry weight basis Sulphates Not more than 100 mg/kg expressed on dry weight basis Nickel Not more than 2 mg/kg expressed on dry weight basis Arsenic Not more than 3 mg/kg expressed on dry weight basis Not more than 1 mg/kg expressed on dry weight basis Lead Heavy metals Not more than 10 mg/kg expressed as Pb on dry weight basis

▼ M3

E 421 MANNITOL

1.	Mannitol	
Syr	nonyms	D-mann

D-mannitol Definition

Manufactured by catalytic hydrogenation of carbohydrate

solutions containing glucose and/or fructose

Chemical name D-mannitol Einecs 200-711-8 Chemical formula $C_6H_{14}O_6$ Molecular weight 182,2

Content not less than 96,0 % D-mannitol and not more than Assay

102 % on the dried basis

Description White, odourless, crystalline powder

1

Identification

A. Solubility

Soluble in water, very slightly soluble in ethanol, practically

insoluble in ether

B. Melting range Between 164 and 169 °C

C. Thin layer chromatography

Passes test

D. Specific rotation

[α] 20 D: + 23° to + 25° (borate solution)

E. pH Between 5 and 8

> Add 0,5 ml of a saturated solution of potassium chloride to 10 ml of a 10 % w/v solution of the sample, then measure the

pН

Purity

Not more than 0,3 % (105 °C, four hours) Loss on drying

Not more than 0,3 % (as glucose) Reducing sugars Total sugars Not more than 1 % (as glucose)

Sulphated ash Not more than 0,1 % Chlorides Not more than 70 mg/kg Sulphate Not more than 100 mg/kg Nickel Not more than 2 mg/kg Lead Not more than 1 mg/kg

Mannitol manufactured by fermentation

Synonyms D-mannitol

Definition Manufactured by discontinous fermentation under aerobic conditions using a conventional strain of the yeast Zygosaccharomyces rouxii Chemical name D-mannitol Einecs 200-711-8 Chemical formula $\mathrm{C_6H_{14}O_6}$ Molecular weight 182.2 Assay Not less than 99 % on the dried basis Description White, odourless crystalline powder Identification A. Solubility Soluble in water, very slightly soluble in ethanol, practically insoluble in ether Between 164 and 169 °C B. Melting range C. Thin layer chromatography passes test D. Specific rotation [α] 20 D: + 23° to + 25° (borate solution) E. pH Between 5 and 8 Add 0,5 ml of a saturated solution of potassium chloride to 10 ml of a 10 % w/v solution of the sample, then measure the

Purity

Arabitol Not more than 0,3 %

Loss on drying Not more than 0,3 % (105 °C, four hours)

Reducing sugars

Not more than 0,3 % (as glucose)

Total sugars

Not more than 1 % (as glucose)

Sulphated ash Not more than 0,1 % Chlorides Not more than 70 mg/kg Sulphate Not more than 100 mg/kg Lead Not more than 1 mg/kg Aerobic mesophilic bacteria Not more than $10^3/g$ Coliforms Absent in 10 g Salmonella Absent in 10 g E. coli Absent in 10 g Staphylococcus aureus Absent in 10 g

Pseudomonas aeruginosa

Moulds

Not more than 100/g

Yeasts

Not more than 100/g

▼M1

E 953 — ISOMALT

Synonyms Hydrogenated isomaltulose, hydrogenated palatinose.

Definition

Chemical name Isomalt is a mixture of hydrogenated mono- and disaccharides

whose principal components are the disaccharides:

6-O-α-D-Glucopyranosyl-D-sorbitol (1,6-GPS) and 1-O-α-D-

Glucopyranosyl-D-mannitol dihydrate (1,1-GPM)

Chemical formula 6-O-α-D-Glucopyranosyl-D-sorbitol: C₁₂H₂₄O₁₁

1-O-α-D-Glucopyranosyl-D-mannitol dihydrate:

 $C_{12}H_{24}O_{11}.2H_2O$

Relative molecular mass 6-O-α-D-Glucopyranosyl-D-sorbitol: 344,32

1-O-α-D-Glucopyranosyl-D-mannitol dihydrate: 380,32

Assay Content not less than 98 % of hydrogenated mono- and disac-

charides and not less than 86 % of the mixture of 6-O- α -D-Glucopyranosyl-D-sorbitol and 1-O- α -D-Glucopyranosyl-D-mannitol dihydrate determined on the anhydrous basis.

Description Odourless, white, slightly hygroscopic, crystalline mass.

▼M1

Identification

A. Solubility

Soluble in water, very slightly soluble in ethanol.

B. Thin layer chromatography

Examine by thin layer chromatography using a plate coated with an approximately 0,2 mm layer of chromatographic silica gel. The principal spots in the chromatogram are those of 1,1-GPM and 1,6-GPS.

Purity

Water content Not more than 7 % (Karl Fischer Method)

Sulphated ash Not more than 0,05 % expressed on the dry weight basis

D-Mannitol Not more than 3 % Not more than 6 %

Reducing sugars Not more than 0,3 % expressed as glucose on the dry weight

basis

Not more than 2 mg/kg expressed on the dry weight basis

Arsenic

Not more than 3 mg/kg expressed on the dry weight basis

Lead

Not more than 1 mg/kg expressed on the dry weight basis

Heavy metals (as Pb)

Not more than 10 mg/kg expressed on the dry weight basis.

▼<u>M5</u>

E 965 (i) MALTITOL

Synonyms

Definition

Chemical name (α)-D-Glucopyranosyl-1,4-D-glucitol

Einecs 209-567-0 Chemical formula $C_{12}H_{24}O_{11}$ Relative molecular mass 344,31

Assay Content not less than 98 % of D-maltitol

 $C_{12}H_{24}O_{11}$ on the anhydrous basis

D-Maltitol, hydrogenated maltose

Description Sweet tasting, white crystalline powder

Identification

A. Solubility Very soluble in water, slightly soluble in ethanol

B. Melting range 148 to 151 °C

C. Specific rotation $[\alpha]_{D^{20}} = +105.5^{\circ} \text{ to } +108.5^{\circ} \text{ (5 \% w/v solution)}$

Purity

Water Not more than 1 % (Karl Fischer method)

Sulphated ash Not more than 0,1 % expressed on dry weight basis

Reducing sugars

Not more than 0,1 % expressed as glucose on dry weight basis

Not more than 50 mg/kg expressed on dry weight basis

Not more than 100 mg/kg expressed on dry weight basis

Nickel

Not more than 2 mg/kg expressed on dry weight basis

Arsenic

Not more than 3 mg/kg expressed on dry weight basis

Lead

Not more than 1 mg/kg expressed on dry weight basis

E 965 (ii) MALTITOL SYRUP

Synonyms

Hydrogenated high-maltose glucose syrup, hydrogenated glucose syrup

Definition

A mixture consisting of mainly maltitol with sorbitol and hydrogenated oligo- and polysaccharides. It is manufactured by the catalytic hydrogenation of high maltose-content glucose syrup or by the hydrogenation of its individual components followed by blending. The article of commerce is supplied both as a syrup and as a solid product

Content not less than 99 % of total hydrogenated saccharides Assay

on the anhydrous basis and not less than 50 % of maltitol on

the anhydrous basis

Colourless and odourless, clear viscous liquids or white crys-Description

talline masses

Identification

A. Solubility Very soluble in water, slightly soluble in ethanol

B. Thin layer chromatography

Passes test

Purity

Water Not more than 31 % (Karl Fischer) Reducing sugars Not more than 0,3 % (as glucose)

Sulphated ash Not more than 0,1 % Chlorides Not more than 50 mg/kg Sulphate Not more than 100 mg/kg Nickel Not more than 2 mg/kg Lead Not more than 1 mg/kg

E 966 LACTITOL

Synonyms Lactit, lactositol, lactobiosit

Definition

 $\hbox{4-O-$\beta$-D-Galactopy ranosyl-D-glucitol}\\$ Chemical name

Einecs 209-566-5 Chemical formula $C_{12}H_{24}O_{11}$ Relative molecular mass 344.32

Not less than 95 % on the dry weight basis Assav

Description Sweet-tasting crystalline powders or colourless solutions. Crys-

talline products occur in anhydrous, monohydrate and

dihydrate forms

Identification

A. Solubility Very soluble in water

B. Specific rotation $[\alpha]_{D^{20}} = + 13^{\circ}$ to $+ 16^{\circ}$ calculated on the anhydrous basis

(10 % w/v aqueous solution)

Purity

Water Crystalline products; not more than 10,5 % (Karl Fischer

method)

Other polyols Not more than 2,5 % on the anhydrous basis

Reducing sugars Not more than 0,2 % expressed as glucose on dry weight basis Chlorides Not more than 100 mg/kg expressed on dry weight basis Sulphates Not more than 200 mg/kg expressed on dry weight basis Sulphated ash Not more than 0,1 % expressed on dry weight basis Nickel Not more than 2 mg/kg expressed on dry weight basis Not more than 3 mg/kg expressed on dry weight basis Arsenic Not more than 1 mg/kg expressed on dry weight basis Lead

▼B

E 967 — XYLITOL

Synonyms **Xylitol**

Definition

Chemical name D-xylitol 201-788-0 Einecs E number E 967 Chemical formula $C_5H_{12}O_5$ Relative molecular mass 152,15

Not less than 98,5% as xylitol on the anhydrous basis Assay

▼B

Description White, crystalline powder, practically odourless with a very

sweet taste

Identification

A. Solubility Very soluble in water, sparingly soluble in ethanol

B. Melting range 92 to 96°C

5 to 7 (10% w/v aqueous solution) C. pH

Purity

Not more than 0,5%. Dry 0,5 g of sample in a vacuum over Loss on drying

phosphorus at 60°C for four hours

Sulphated ash Not more than 0,1% expressed on dry weight basis

Not more than 0,2% expressed as glucose on dry weight basis Reducing sugars

Other polyhydric alcohols Not more than 1% expressed on dry weight basis

Nickel Not more than 2 mg/kg expressed on dry weight basis Arsenic Not more than 3 mg/kg expressed on dry weight basis Lead Not more than 1 mg/kg expressed on dry weight basis Heavy metals Not more than 10 mg/kg expressed as Pb on dry weight basis

Chlorides Not more than 100 mg/kg expressed on dry weight basis

Not more than 200 mg/kg expressed on dry weight basis Sulphates

▼ M5

E 968 ERYTHRITOL

Synonyms Meso-erythritol, tetrahydroxybutane, erythrite

Definition Obtained by fermentation of carbohydrate source by safe and

> suitable food grade osmophilic yeasts such as Moniliella pollinis or Trichosporonoides megachilensis, followed by puri-

fication and drying

Chemical name 1,2,3,4-Butanetetrol

Einecs 205-737-3 Chemical formula $C_4H_{10}O_4$ Molecular weight 122,12

Not less than 99 % after drying Assay

Description White, odourless, non-hygroscopic, heat-stable crystals with a

sweetness of approximately 60-80 % that of sucrose.

Identification

A. Solubility Freely soluble in water, slightly soluble in ethanol, insoluble in

diethyl ether.

119-123 °C B. Melting range

Purity

Not more than 0,2 % (70 °C, six hours, in a vacuum Loss on drying

desiccator)

Sulphated ash Not more than 0,1 %

Reducing substances Not more than 0,3 % expressed as D-glucose

Ribitol and glycerol Not more than 0,1 % Lead Not more than 0,5 mg/kg

▼ M3

E 950 — ACESULFAME K

Acesulfame potassium, potassium salt of 3,4-dihydro-6-methyl-**Synonyms**

1,2,3-oxathiazin-4-one, 2,2-dioxide

Definition

Chemical name 6-methyl-1,2,3-oxathiazin-4(3H)-one-2,2-dioxide potassium salt

259-715-3 Einecs Chemical formula C₄H₄KNO₄S

201,24 Molecular weight

Assay Content not less than 99 % of C₄H₄KNO₄S on the anhydrous

basis

Description Odourless, white, crystalline powder. Approximately 200 times

as sweet as sucrose

Identification

A. Solubility Very soluble in water, very slightly soluble in ethanol

B. Ultra violet absorption Maximum 227 ± 2 nm for a solution of 10 mg in 1 000 ml of

water

C. Positive test for potassium Passes test (test the residue obtained by igniting 2 g of the

sample)

D. Precipitation test

Add a few drops of a 10 % solution of sodium cobaltnitrite to a solution of 0,2 g of the sample in 2 ml of acetic acid and

2 ml of water. A yellow precipitate is produced

Purity

Loss on drying Not more than 1 % (105 °C, two hours)

Organic impurities Passes test for 20 mg/kg of UV active components

Fluoride Not more than 3 mg/kg
Lead Not more than 1 mg/kg

▼<u>B</u>

E 951 — ASPARTAME

Synonyms Aspartyl phenylalanine methyl ester

Definition

Chemical name N-L-α-(Aspartyl-L-phenylalanine-1-methyl ester, 3-amino-N-

 $(\alpha\text{-carbomethoxy-phenethyl})\text{-succinamic acid-N-methyl ester}$

Relative molecular mass 294,31

Assay Not less than 98% and not more than 102% of C₁₄H₁₈N₂O₅ on

the anhydrous basis

Description White, odourless, crystalline powder having a sweet taste.

Approximately 200 times as sweet as sucrose

Identification

Solubility Slightly soluble in water and in ethanol

Purity

Loss on drying Not more than 4,5% (105°C, four hours)

Sulphated ash Not more than 0,2% expressed on dry weight basis

pH Between 4,5 and 6,0 (1 in 125 solution)

Transmittance of a 1% solution in 2N hydrochloric acid,

determined in a 1-cm cell at 430 nm with a suitable spectrophotometer, using 2N hydrochloric acid as a reference, is not less than 0,95, equivalent to an absorbance of not more than

approximately 0,022

Specific rotation $(\alpha)_D^{20}$: +14,5 to +16,5°

Determine in a 4 in 100/15 N formic acid solution within 30

minutes after preparation of the sample solution

Arsenic Not more than 3 mg/kg expressed on dry weight basis

Lead Not more than 1 mg/kg expressed on dry weight basis

Heavy metals Not more than 10 mg/kg expressed as Pb on dry weight basis

5-Benzyl-3,6-dioxo-2-piperazi- Not more than 1,5% expressed on dry weight basis

neacetic acid

E 952 — CYCLAMIC ACID AND ITS Na AND Ca SALTS

(I) CYCLAMIC ACID

Synonyms

Cyclohexylsulphamic acid, cyclamate

Definition

Chemical name Cyclohexanesulphamic acid, cyclohexylaminosulphonic acid

Einecs202-898-1E numberE 952Chemical formula $C_6H_{13}NO_3S$ Relative molecular mass179,24

Assay Cyclohexylsulphamic acid contains not less than 98% and not more than the equivalent of 102% of $C_6H_{13}NO_3S$, calculated

on the anhydrous basis

Description A practically colourless, white crystalline powder with a sweet-

sour taste. Approximately 40 times as sweet as sucrose

Identification

A. Solubility Soluble in water and in ethanol

approximately molar solution of barium chloride in water and filter if any haze or precipitate forms. To the clear solution add 1 ml of a 10% solution of sodium nitrite. A white precipitate

forms.

Purity

Loss on drying Not more than 1% (105°C, one hour)

Selenium Not more than 30 mg/kg expressed as selenium on dry weight

basis

Lead Not more than 1 mg/kg expressed on dry weight basis

Heavy metals Not more than 10 mg/kg expressed as Pb on dry weight basis

Arsenic Not more than 3 mg/kg expressed on dry weight basis

Cyclohexylamine Not more than 10 mg/kg expressed on dry weight basis

Dicyclohexylamine Not more than 1 mg/kg expressed on dry weight basis

Aniline Not more than 1 mg/kg expressed on dry weight basis

(II) SODIUM CYCLAMATE

Synonyms Cyclamate, sodium salt of cyclamic acid

Definition

Chemical name Sodium cyclohexanesulphamate, sodium cyclohexylsulphamate

 Einecs
 205-348-9

 E number
 E 952

Chemical formula C₆H₁₂NNaO₃S and the dihydrate form C₆H₁₂NNaO₃S·2H₂O

Relative molecular mass 201,22 calculated on the anhydrous form 237,22 calculated on the hydrated form

Assay

Not les sthan 98% and not more than 102% on the dried basis

Dihydrate form: not less than 84% on the dried basis

DescriptionWhite, odourless crystals or crystalline powder. Approximately 30 times as sweet as sucrose

Identification

Soluble in water, practically insoluble in ethanol

Purity

Loss on drying Not more than 1% (105°C, one hour)

Not more than 15,2% (105°C, two hours) for the dihydrate

form

Selenium Not more than 30 mg/kg expressed as selenium on dry weight

basis

Arsenic Not more than 3 mg/kg expressed on dry weight basis

Lead Not more than 1 mg/kg expressed on dry weight basis

Heavy metals Not more than 10 mg/kg expressed as Pb on dry weight basis

Cyclohexylamine Not more than 10 mg/kg expressed on dry weight basis

Dicyclohexylamine Not more than 1 mg/kg expressed on dry weight basis

Aniline Not more than 1 mg/kg expressed on dry weight basis

▼<u>B</u>

(III) CALCIUM CYCLAMATE

Synonyms Cyclamate, calcium salt of cyclamic acid

Definition

Chemical name Calcium cyclohexanesulphamate, calcium cyclohexylsul-

phamate

 Einecs
 205-349-4

 E number
 E 952

Chemical formula $C_{12}H_{24}CaN_2O_6S_2 \cdot 2H_2O$

Relative molecular mass 432,57

Assay Not less than 98% and not more than 10% on the dried basis

Description White, colourless crystals or crystaline powder. Approximately

30 times as sweet as sucrose

Identification

Soluble in water, sparingly soluble in ethanol

Purity

Loss on drying Not more than 1% (105°C, one hour)

Not more than 8,5% (140°C, four hours) for the dihydrate form

Selenium Not more than 30 mg/kg expressed as selenium on dry weight

basis

Arsenic Not more than 3 mg/kg expressed on dry weight basis

Lead Not more than 1 mg/kg expressed on dry weight basis

Heavy metals Not more than 10 mg/kg expressed as Pb on dry weight basis

Cyclohexylamine

Not more than 10 mg/kg expressed on dry weight basis

Not more than 1 mg/kg expressed on dry weight basis

Aniline

Not more than 1 mg/kg expressed on dry weight basis

▼<u>M5</u>

E 954 SACCHARIN AND ITS Na, K AND Ca SALTS

(I) SACCHARIN

Definition

Chemical name 3-Oxo-2,3-dihydrobenzo(d)isothiazol-1,1-dioxide

Einecs 201-321-0 Chemical formula $C_7H_5NO_3S$ Relative molecular mass 183,18

Assay Not less than 99 % and not more than 101 % of C₇H₅NO₃S on

the anhydrous basis

Description White crystals or a white crystalline powder, odourless or with

a faint, aromatic odour, having a sweet taste, even in very dilute solutions. Approximately between 300 and 500 times

as sweet as sucrose

Identification

Solubility Slightly soluble in water, soluble in basic solutions, sparingly

soluble in ethanol

Purity

Loss on drying Not more than 1 % (105 °C, two hours)

Melting range 226-230 °C

Sulphated ash Not more than 0,2 % expressed on dry weight basis

Benzoic and salicylic acid To 10 ml of a 1 in 20 solution, previously acidified with five

drops of acetic acid, add three drops of an approximately molar solution of ferric chloride in water. No precipitate or violet

colour appears

o-Toluenesulphonamide
Pot more than 10 mg/kg expressed on dry weight basis
Not more than 10 mg/kg expressed on dry weight basis
Benzoic acid p-sulfonamide
Not more than 10 mg/kg expressed on dry weight basis
Benzoic acid p-sulfonamide

Readily carbonisable

substances

Absent

Arsenic Selenium Lead Not more than 3 mg/kg expressed on dry weight basis Not more than 30 mg/kg expressed on dry weight basis Not more than 1 mg/kg expressed on dry weight basis.

(II) SODIUM SACCHARIN

Synonyms

Saccharin, sodium salt of saccharin

Definition

Chemical name

Sodium o-benzosulphimide, sodium salt of 2,3-dihydro-3-oxobenzisosulphonazole, oxobenzisosulphonazole, 1,2-benzisothiazolin-3-one-1,1-dioxide sodium salt dihydrate

Einecs 204-886-1

Chemical formula C₇H₄NNaO₃S·2H₂O

Relative molecular mass 241,19

Assay Not less than 99 % and not more than 101 % of C₇H₄NNaO₃S

on the anhydrous basis

Description White crystals or a white crystalline efflorescent powder,

odourless or with a faint odour, having an intensely sweet taste, even in very dilute solutions. Approximately between 300 and 500 times as sweet as sucrose in dilute solutions

Identification

Solubility Freely soluble in water, sparingly soluble in ethanol

Purity

Loss on drying Not more than 15 % (120 °C, four hours)

Benzoic and salicylic acid To 10 ml of a 1 in 20 solution, previously acidified with five

drops of acetic acid, add three drops of an approximately molar solution of ferric chloride in water. No precipitate or violet colour appears

colour appears

o-Toluenesulphonamide Not more than 10 mg/kg expressed on dry weight basis p-Toluenesulphonamide Not more than 10 mg/kg expressed on dry weight basis

Benzoic acid p-sulphonom Not more than 25 mg/kg expressed on dry weight basis namide

Readily carbonisable

Readily carbonisable substances

Arsenic Not more than 3 mg/kg expressed on dry weight basis

Selenium Not more than 30 mg/kg expressed on dry weight basis

Lead Not more than 1 mg/kg expressed on dry weight basis

(III) CALCIUM SACCHARIN

Synonyms Saccharin, calcium salt of saccharin

Absent

Definition

Chemical name Calcium o-benzosulphimide, calcium salt of 2,3-dihydro-3-

oxobenzisosulfonazole, 1,2-benzisothiazolin-3-one-1,1-dioxide

calcium salt hydrate (2:7)

Einecs 229-349-9

Chemical formula $C_{14}H_8CaN_2O_6S_2\cdot 3^1/_2H_2O$

Relative molecular mass 467,48

Assay Not less than 95 % of C₁₄H₈CaN₂O₆S₂ on the anhydrous basis

Description White crystals or a white crystalline powder, odourless or with

a faint odour, having an intensely sweet taste, even in very dilute solutions. Approximately between 300 and 500 times as

sweet as sucrose in dilute solutions

Identification

Solubility Freely soluble in water, soluble in ethanol

Purity Loss on drying Not more than 13,5 % (120 °C, four hours) Benzoic and salicylic acid To 10 ml of a 1 in 20 solution, previously acidified with five drops of acetic acid, add three drops of an approximately molar solution of ferric chloride in water. No precipitate or violet o-Toluenesulphonamide Not more than 10 mg/kg expressed on dry weight basis p-Toluenesulphonamide Not more than 10 mg/kg expressed on dry weight basis Benzoic acid p-sulpho-Not more than 25 mg/kg expressed on dry weight basis namide Readily carbonisable Absent substances Arsenic Not more than 3 mg/kg expressed on dry weight basis Not more than 30 mg/kg expressed on dry weight basis Selenium Lead Not more than 1 mg/kg expressed on dry weight basis (IV)POTASSIUM SACCHARIN **Synonyms** Saccharin, potassium salt of saccharin **Definition** Chemical name Potassium o-benzosulphimide, potassium salt of 2,3-dihydro-3oxobenzisosulphonazole, potassium salt of 1,2-benzisothiazolin-3-one-1,1-dioxide monohydrate Einecs Chemical formula C7H4KNO3S·H2O Relative molecular mass 239,77 Assay Not less than 99 % and not more than 101 % of C7H4KNO3S on the anhydrous basis Description White crystals or a white crystalline powder, odourless or with a faint odour, having an intensely sweet taste, even in very dilute solutions. Approximately between 300 and 500 times as sweet as sucrose Identification Solubility Freely soluble in water, sparingly soluble in ethanol **Purity** Loss on drying Not more than 8 % (120 °C, four hours) Benzoic and salicylic acid To 10 ml of a 1 in 20 solution, previously acidified with five drops of acetic acid, add three drops of an approximately molar solution of ferric chloride in water. No precipitate or violet colour appears o-Toluenesulphonamide Not more than 10 mg/kg expressed on dry weight basis p-Toluenesulphonamide Not more than 10 mg/kg expressed on dry weight basis Benzoic acid p-sulpho-Not more than 25 mg/kg expressed on dry weight basis namide Readily carbonisable Absent substances Arsenic Not more than 3 mg/kg expressed on dry weight basis Selenium Not more than 30 mg/kg expressed on dry weight basis

E 955 SUCRALOSE

Lead

Synonyms 4,1',6'-Trichlorogalactosucrose

Definition

Chilition

Chemical name 1,6-Dichloro-1,6-dideoxy-β-D-fructofuranosyl-4-chloro-4-

Not more than 1 mg/kg expressed on dry weight basis

deoxy-α-D-galactopyranoside

Einecs 259-952-2 Chemical formula $C_{12}H_{19}Cl_3O_8$

Molecular weight 397,64

Assay Content not less than 98 % and not more than 102 % of

 $C_{12}H_{19}Cl_3O_8$ calculated on an anhydrous basis.

Description White to off-white, practically odourless crystalline powder.

Identification

A. Solubility Freely soluble in water, methanol and ethanol

Slightly soluble in ethyl acetate

B. Infrared absorption The infrared spectrum of a potassium bromide dispersion of the

sample exhibits relative maxima at similar wave numbers as those shown in the reference spectrum obtained using a

sucralose reference standard

C. Thin layer chromatography | The main spot in the test solution has the same Rf value as that

of the main spot of standard solution A referred to in the test for other chlorinated disaccharides. This standard solution is obtained by dissolving 1,0 g of sucralose reference standard

in 10 ml of methanol

D. Specific rotation $[\alpha]_{D^{20}} = +84.0^{\circ}$ to $+87.5^{\circ}$ calculated on the anhydrous basis

(10 % w/v solution)

Purity

Water Not more than 2,0 % (Karl Fischer method)

Sulphated ash
Other chlorinated disaccharides
Chlorinated monosaccharides
Triphenylphosphine oxide
Methanol

Not more than 0,7 %
Not more than 0,5 %
Not more than 0,1 %
Not more than 150 mg/kg
Not more than 0,1 %

Lead Not more than 1 mg/kg

▼<u>B</u>

E 957 — THAUMATIN

Synonyms

Definition

Chemical name Thaumatin is obtained by aqueous extraction (pH 2,5 to 4) of

the arils of the fruit of the natural strain of *Thaumatococcus daniellii* (Benth) and consists essentially of the proteins thaumatin I and thaumatin II together with minor amounts of

plant constituents derived from the source material

 Einecs
 258-822-2

 E number
 E 957

Chemical formula Polypeptide of 207 aminoacids

Relative molecular mass Thaumatin I 22209
Thaumatin II 22293

Assay Not less than 16% nitrogen on the dried basis equivalent to not

less than 94% proteins $(N \times 5,8)$

Description Odourless, cream-coloured powder with an intensely sweet

taste. Approximately 2 000 to 3 000 times as sweet as sucrose

Identification

Solubility Very soluble in water, insoluble in acetone

Purity

Loss on drying Not more than 9% (105°C to constant weight)

Carbohydrates Not more than 3% expressed on dry weight basis

 Sulphated ash
 Not more than 2% expressed on dry weight basis

 Aluminium
 Not more than 100 mg/kg expressed on dry weight basis

 Arsenic
 Not more than 3 mg/kg expressed on dry weight basis

Lead 3 mg/kg expressed on dry weight basis

Microbiological criteria Total aerobic microbial count: Max 1 000/g E. Coli: absent in

1 g

E 959 — NEOHESPERIDINE DIHYDROCHALCONE

Synonyms Neohesperidin dihydrochalcone, NHDC, hesperetin dihydrochalcone-4'-β-neohesperidoside, neohesperidin DC

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Definition

Chemical name 2-O-α-L-rhamnopyranosyl-4'-β-D-glucopyranosyl hesperetin

dihydrochalcone; obtained by catalytic hydrogenation of neohe-

speridin

Einecs243-978-6E numberE 959Chemical formula $C_{28}H_{36}O_{15}$ Relative molecular mass612,6

Assay Content not less than 96% on the dried basis

Description Off white, odourless, crystalline powder having a characteristic,

intensive sweet taste. Approximately between 1 000 and 1 800

times as sweet as sucrose

Identification

A. Solubility Freely soluble in hot water, very slightly soluble in cold water,

practically insoluble in ether and benzene

B. Ultraviolet absorption

maximum

282 to 283 nm for a solution of 2 mg in 100 ml methanol

C. Neu's test

Dissolve about 10 mg of neohesperidine DC in 1 ml methanol,

add 1 ml of a 1% 2-aminoethyl diphenyl borate methanolic

solution. A bright yellow colour is produced

Purity

Loss on drying Not more than 11% (105°C, three huors)

Sulphated ash

Not more than 0,2% expressed on dry weight basis

Arsenic

Not more than 3 mg/kg expressed on dry weight basis

Lead

Not more than 2 mg/kg expressed on dry weight basis

Heavy metals Not more than 10 mg/kg expressed as Pb on dry weight basis

▼ M5

E 962 SALT OF ASPARTAME-ACESULFAME

Synonyms Aspartame-acesulfame, aspartame-acesulfame salt

Definition The salt is prepared by heating an approximately 2:1 ratio (w/

w) of aspartame and acesulfame K in solution at acidic pH and allowing crystallisation to occur. The potassium and moisture are eliminated. The product is more stable than aspartame

alone

Chemical name 6-Methyl-1,2,3-oxathiazine-4(3H)-one-2,2-dioxide salt of L-

phenylalanyl-2-methyl-L-α-aspartic acid

Chemical formula $C_{18}H_{23}O_9N_3S$ Molecular weight 457,46

Assay 63,0 % to 66,0 % aspartame (dry basis) and 34,0 % to 37 %

acesulfame (acid form on a dry basis)

Description A white, odourless, crystalline powder

Identification

A. Solubility Sparingly soluble in water, slightly soluble in ethanol

B. Transmittance The transmittance of a 1 % solution in water determined in a 1

cm cell at 430 nm with a suitable spectrophotometer using water as a reference, is not less than 0,95, equivalent to an

absorbance of not more than approximately 0,022

C. Specific rotation $\left[\alpha\right]_{D^{20}} = +14.5^{\circ} \text{ to } +16.5^{\circ}$

Determine at concentration of 6,2 g in 100 ml formic acid (15N) within 30 min of preparation of the solution. Divide the calculated specific rotation by 0,646 to correct for the aspartame content of the salt of aspartame-acesulfame

▼<u>M5</u>

Purity

Loss on drying

5-Benzyl-3,6-dioxo-2-piperazineacetic acid

Lead

Not more than 0,5 % (105 °C, four hours)

Not more than 0,5 %

Not more than 1 mg/kg