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COMMISSION DIRECTIVE 95/31/EC

of 5 July 1995

laying down specific criteria of purity concerning sweeteners for use in foodstuffs (Text with EEA relevance)

(OJ L 178, 28.7.1995, p. 1)

Amended by:

<u>B</u>

		Official Journal		
		No	page	date
<u>M1</u>	Commission Directive 98/66/EC of 4 September 1998	L 257	35	19.9.1998
► <u>M2</u>	Commission Directive 2000/51/EC of 26 July 2000	L 198	41	4.8.2000
► <u>M3</u>	Commission Directive 2001/52/EC of 3 July 2001	L 190	18	12.7.2001
<u>M4</u>	Commission Directive 2004/46/EC of 16 April 2004	L 114	15	21.4.2004

COMMISSION DIRECTIVE 95/31/EC

of 5 July 1995

laying down specific criteria of purity concerning sweeteners for use in foodstuffs

(Text with EEA relevance)

THE COMMISSION OF THE EUROPEAN COMMUNITIES,

Having regard to the Treaty establishing the European Community,

Having regard to Council Directive 89/107/EEC of 21 December 1988 on the approximation of the laws of the Member States concerning food additives authorized for use in foodstuffs intended for human consumption (1), as amended by Directive 94/34/EC (2), and in particular Article 3 (3) (a) thereof,

After consultation of the Scientific Committee on Food,

Whereas it is necessary to establish purity criteria for all sweeteners mentioned in European Parliament and Council Directive 94/35/EC of 30 June 1994 on sweeteners for use in foodstuffs (3);

Whereas it is necessary to take into account the specifications and analytical techniques for sweeteners as set out in the *Codex Alimentarius* and the Joint FAO/WHO Expert Committee on Food Additives (Jecfa);

Whereas food additives, prepared by production methods or starting materials significantly different from those included in the evaluation of the Scientific Committee for Food, or different from those mentioned in this Directive, should be submitted for evaluation by the Scientific Committee for Food with a view to full evaluation with emphasis on the purity criteria;

Whereas the measures provided for in this Directive are in line with the opinion of the Standing Committee on Foodstuffs,

HAS ADOPTED THIS DIRECTIVE:

Article 1

- 1. Purity criteria mentioned under Article 3 (3) (a) of Directive 89/107/EEC for sweeteners mentioned in Directive 94/35/EC are set out in the Annex.
- 2. The purity criteria for E 420 (i), E 420 (ii) and E 421 mentioned in the Annex to this Directive supersede the purity criteria for the said substances mentioned in the Annex to Council Directive 78/663/ EEC (4).

Article 2

1. Member States shall bring into force the laws, regulations and administrative provisions necessary to comply with this Directive not later than 1 July 1996. They shall forthwith inform the Commission thereof.

When Member States adopt these provisions, these shall contain a reference to this Directive or shall be accompanied by such reference at the time of their official publication. The procedure for such reference shall be adopted by Member States.

2. Products put on the market or labelled before that date which do not comply with this Directive may, however, be marketed until stocks are exhausted.

⁽¹⁾ OJ No L 40, 11. 2. 1989, p. 27.

⁽²⁾ OJ No L 237, 10. 9. 1994, p. 1.

⁽³⁾ OJ No L 237, 10. 9. 1994, p. 3.

⁽⁴⁾ OJ No L 223, 14. 8. 1978, p. 7.

Article 3

This Directive shall enter into force on the 20th day following its publication in the *Official Journal of the European Communities*.

Article 4

This Directive is addressed to the Member States.

ANNEX

E 420 (i) — SORBITOL

Synonyms D-glucitol, D-sorbitol

Definition

Chemical nameD-glucitolEinecs200-061-5E numberE 420 (i)Chemical formula $C_6H_{14}O_6$ Relative molecular mass182,17

Assay Content not less than 97 % of total glycitols and not less

than 91 % of D-sorbitol on the dry weight basis.

Glycitols are compounds with the structural formula CH_2OH - $(CHOH)_n$ - CH_2OH , where 'n' is an integer

DescriptionWhite hygroscopic powder, crystalline powder, flakes or granules having a sweet taste

Identification

A. Solubility Very soluble in water, slightly soluble in ethanol

B. Melting range 88 to 102 °C

C. Sorbitol monobenzylidene derivative To 5 g of the sample add 7 ml of methanol, 1 ml of benzaldehyde and 1 ml of hydrochloric acid. Mix and shake in a mechanical shaker until crystals appear. Filter with the aid of suction, dissolve the crystals in 20 ml of boiling water containing 1 g of sodium bicarbonate, filter while hot, cool the filtrate, filter with suction, wash with 5 ml of methanol-water mixture (1 in 2) and dry in air. The crystals so obtained melt between 173 and 179 °C

Purity

Water content Not more than 1 % (Karl Fischer method)

Sulphated ash Not more than 0,1 % expressed on dry weight basis

Reducing sugars Not more than 0,3 % expressed as glucose on dry

weight basis

Total sugars Not more than 1 % expressed as glucose on dry weight

hasis

Chlorides Not more than 50 mg/kg expressed on dry weight basis

Sulphates Not more than 100 mg/kg expressed on dry weight basis

Nickel Not more than 2 mg/kg expressed on dry weight basis

Arsenic Not more than 3 mg/kg expressed on dry weight basis

Lead Not more than 1 mg/kg expressed on dry weight basis

Heavy metals Not more than 10 mg/kg expressed as Pb on dry weight

basis

E 420 (ii) — SORBITOL SYRUP

Synonyms D-clucitol syrup

Definition

Chemical name

Sorbitol syrup formed by hydrogenation of glucose syrup is composed of D-sorbitol, D-mannitol and hydrogenated saccharides.

The part of the product which is not D-sorbitol is composed mainly of hydrogenated oligosaccharides formed by the hydrogenation of glucose syrup used as raw material (in which case the syrup is non-crystallizing) or mannitol. Minor quantities of glycitols where

 $n \le 4$ may be present. Glycitols are compounds with the structural formula CH₂OH-(CHOH)_n-CH₂OH, where 'n' is an integer

270-337-8

E 420 (ii) E number

Assay Content not less than 69 % total solids and not less than

50 % of D-sorbitol on the anhydrous basis

Description Clear colourless and sweet tasting aqueous solution

Identification

Einecs

A. Solubility Miscible with water, with glycerol, and with propane-

To 5 g of the sample add 7 ml of methanol, 1 ml of B. Sorbitol monobenzylidene deriva-

benzaldehyde and 1 ml of hydrochloric acid. Mix and shake in a mechanical shaker until crystals appear. Filter with the aid of suction, dissolve the crystals in 20 ml of boiling water containing 1 g of sodium bicarbonate, filter while hot. Cool the filtrate filter with suction, wash with 5 ml of methanol-water mixture (1 in 2) and dry in air. The crystals so obtained melt between 173

and 179 °C

Purity

Water content Not more than 31 % (Karl Fischer method)

Sulphated ash Not more than 0,1 % expressed on dry weight basis

Reducing sugars Not more than 0,3 % expressed as glucose on dry

weight basis

Chlorides Not more than 50 mg/kg expressed on dry weight basis Sulphates Not more than 100 mg/kg expressed on dry weight basis Nickel Not more than 2 mg/kg expressed on dry weight basis Arsenic Not more than 3 mg/kg expressed on dry weight basis

Lead Not more than 1 mg/kg expressed on dry weight basis

Heavy metals Not more than 10 mg/kg expressed as Pb on dry weight

▼M3

E 421 MANNITOL

Synonyms

1. Mannitol	

D-mannitol Definition Manufactured by catalytic hydrogenation of carbohy-

drate solutions containing glucose and/or fructose

Chemical name D-mannitol 200-711-8 Einecs

Chemical formula $C_{6}H_{14}O_{6}$ Molecular weight 182,2

Content not less than 96,0 % D-mannitol and not more Assay

than 102 % on the dried basis

Description White, odourless, crystalline powder

Identification

A. Solubility Soluble in water, very slightly soluble in ethanol, practi-

cally insoluble in ether

B. Melting range Between 164 and 169 °C

C. Thin layer chromatography Passes test

 $[\alpha]^{20}_{p}$: + 23° to + 25° (borate solution) D. Specific rotation

▼<u>M3</u>

E. pH Between 5 and 8

Add 0,5 ml of a saturated solution of potassium chloride to 10 ml of a 10 % w/v solution of the sample, then

measure the pH

Purity

Loss on drying Not more than 0,3 % (105 °C, four hours)

Reducing sugars

Not more than 0,3 % (as glucose)

Total sugars

Not more than 1 % (as glucose)

Sulphated ash

Chlorides

Not more than 0,1 %

Not more than 70 mg/kg

Sulphate

Not more than 100 mg/kg

Not more than 2 mg/kg

Lead

Not more than 1 mg/kg

2 . Mannitol manufactured by fermentation

Synonyms D-mannitol

Definition Manufactured by discontinuous fermentation under

aerobic conditions using a conventional strain of the

yeast Zygosaccharomyces rouxii

Chemical nameD-mannitolEinecs200-711-8Chemical formula $C_6H_{14}O_6$ Molecular weight182,2

Assay Not less than 99 % on the dried basis

Description White, odourless crystalline powder

Identification

A. Solubility Soluble in water, very slightly soluble in ethanol, practi-

cally insoluble in ether

B. Melting range Between 164 and 169 °C

C. Thin layer chromatography passes test

D. Specific rotation $\left[\alpha\right]^{20}_{p}$: + 23° to + 25° (borate solution)

E. pH Between 5 and 8

Add 0,5 ml of a saturated solution of potassium chloride

to 10 ml of a 10 % w/v solution of the sample, then

measure the pH

Purity

Arabitol Not more than 0,3 %

Loss on drying Not more than 0,3 % (105 °C, four hours)

Reducing sugars

Not more than 0,3 % (as glucose)

Total sugars

Not more than 1 % (as glucose)

Sulphated ash

Chlorides

Not more than 0,1 %

Not more than 70 mg/kg

Sulphate

Not more than 100 mg/kg

Lead

Not more than 1 mg/kg

Aerobic mesophilic bacteria

Not more than 10³/g

Coliforms Absent in 10 g

Salmonella Absent in 10 g

E. coli Absent in 10 g

▼<u>M3</u>

Staphylococcus aureus

Absent in 10 g

Pseudomonas aeruginosa

Absent in 10 g

Moulds Yeasts

Not more than 100/g

Not more than 100/g

▼<u>M1</u>

E 953 — ISOMALT

Synonyms

Hydrogenated isomaltulose, hydrogenated palatinose.

Definition

Chemical name

Isomalt is a mixture of hydrogenated mono- and disaccharides whose principal components are disaccharides:

6-O-α-D-Glucopyranosyl-D-sorbitol (1,6-GPS) and 1-O-

α-D-Glucopyranosyl-D-mannitol dihydrate (1,1-GPM)

Chemical formula

6-O-α-D-Glucopyranosyl-D-sorbitol: C₁₂H₂₄O₁₁

 $1\text{-}O\text{-}\alpha\text{-}D\text{-}Glucopyranosyl\text{-}D\text{-}mannitol$

dihydrate:-

C₁₂H₂₄O₁₁.2H₂O

Relative molecular mass

6-O-α-D-Glucopyranosyl-D-sorbitol: 344,32

1-O-α-D-Glucopyranosyl-D-mannitol dihydrate: 380,32

Assay

Content not less than 98 % of hydrogenated mono- and disaccharides and not less than 86 % of the mixture of 6-O-α-D-Glucopyranosyl-D-sorbitol 1-O-α-Dand Glucopyranosyl-D-mannitol dihydrate determined on

the anhydrous basis.

Description

Odourless, white, slightly hygroscopic, crystalline mass.

Identification

A. Solubility

Soluble in water, very slightly soluble in ethanol.

B. Thin layer chromatography

Examine by thin layer chromatography using a plate coated with an approximately 0,2 mm layer of chromatographic silica gel. The principal spots in the chromatogram are those of 1,1-GPM and 1,6-GPS.

Purity

Water content

Not more than 7 % (Karl Fischer Method)

Sulphated ash

Not more than 0,05 % expressed on the dry weight basis

D-Mannitol

Not more than 3 %

D-Sorbitol

Not more than 6 %

Reducing sugars

Not more than 0,3 % expressed as glucose on the dry weight basis

Nickel

Not more than 2 mg/kg expressed on the dry weight basis

Arsenic

Not more than 3 mg/kg expressed on the dry weight

Lead

Not more than 1 mg/kg expressed on the dry weight basis

Heavy metals (as Pb)

Not more than 10 mg/kg expressed on the dry weight

E 965 (i) — MALTITOL

Synonyms D-maltitol, hydrogenated maltose

Definition

Chemical name (α)-D-glucopyranosyl-1,4-D-glucitol

209-567-0 Einecs E number E 965 (i) Chemical formula $C_{12}H_{24}O_{11}$ Relative molecular mass 344,31

Assav Content not less than 98 % D-mannitol C₁₂H₂₄O₁₁ on the

anhydrous basis

Description Sweet tasting, white crystalline powder

Identification

A. Solubility Very soluble in water, slightly soluble in ethanol

148 to 151 °C B. Melting range

 $(\alpha)_D^{20}$ = + 105,5 to + 105,5° (5 % w/v solution) C. Specific rotation

Purity

Water content Not more than 1 % (Karl Fischer method)

Sulphated ash Not more than 0,1 % expressed on dry weight basis

Not more than 0,1 % expressed as glucose on dry Reducing sugars

weight basis

Chlorides Not more than 50 mg/kg expressed on dry weight basis Sulphates Not more than 100 mg/kg expressed on dry weight basis Nickel Not more than 2 mg/kg expressed on dry weight basis Arsenic Not more than 3 mg/kg expressed on dry weight basis Not more than 1 mg/kg expressed on dry weight basis Lead

Not more than 10 mg/kg expressed as Pb on dry weight Heavy metals

basis

▼<u>M2</u>

E 965 (ii) SYRUP MALTITOL

Synonyms Hydrogenated high-maltose-glucose syrup, hydrogenated

glucose syrup

Definition A mixture consisting of mainly maltitol with sorbitol and hydrogenated oligo- and polysaccharides. It is

manufactured by the catalytic hydrogenation of high maltose-content glucose syrup. The article of commerce is supplied both as a syrup and as a solid product.

Assay Content not less than 99 % of total hydrogenated

saccharides on the anhydrous basis and not less than

50 % of maltitol on the anhydrous basis

Description Colourless and odourless, clear viscous liquids or white

crystalline masses

Identification

Very soluble in water, slightly soluble in ethanol A. Solubility

B. Thin layer chromatography Passes test

Purity

Water Not more than 31 % (Karl Fischer) Reducing sugars Not more than 0,3 % (as glucose)

Sulphated ash Not more than 0,1 %

▼<u>M2</u>

Not more than 50 mg/kg Chlorides Sulphate Not more than 100 mg/kg Nickel Not more than 2 mg/kg Not more than 1 mg/kg Lead

▼B

E 966 — LACTITOL

Synonyms Lactit, lactositol, lactobiosit

Definition

Chemical name 4-O-β-D-galactopyranosyl-D-glucitol

Einecs 209-566-5 E number E 966 Chemical formula $C_{12}H_{24}O_{11}$ 344,32 Relative molecular mass

Assay Not less than 95 % on the dry weight basis

Sweet-tasting crystalline powders or colourless solutions. Crystalline products occur in anhydrous,

monohydrate and dihydrate forms

Identification

Description

A. Solubility Very soluble in water

 $(\alpha)_D^{20}=+$ 13 to + 16° calculated on the anhydrous basis (10 % w/v aqueous solution) B. Specific rotation

Purity

Water content Crystalline products; not more than 10,5 % (Karl Fischer

method)

Other polyols Not more than 2,5 % on the anhydrous basis

Reducing sugars Not more than 0,2 % expressed as glucose on dry

weight basis

Chlorides Not more than 100 mg/kg expressed on dry weight basis Sulphates Not more than 200 mg/kg expressed on dry weight basis Sulphated ash Not more than 0,1 % expressed on dry weight basis Nickel Not more than 2 mg/kg expressed on dry weight basis Arsenic Not more than 3 mg/kg expressed on dry weight basis Not more than 1 mg/kg expressed on dry weight basis Lead Heavy metals Not more than 10 mg/kg expressed as Pb on dry weight basis

Xylitol **Synonyms**

Definition

E 967 — XYLITOL

Chemical name D-xylitol Einecs 201-788-0 E 967 E number Chemical formula $C_5H_{12}O_5$ Relative molecular mass 152,15

Not less than 98,5 % as xylitol on the anhydrous basis Assay

Description White, crystalline powder, practically odourless with a

very sweet taste

Identification

A. Solubility Very soluble in water, sparingly soluble in ethanol

B. Melting range 92 to 96 °C

C. pH 5 to 7 (10 % w/v aqueous solution)

Purity

Loss on drying Not more than 0,5 %. Dry 0,5 g of sample in a vacuum

over phosphorus at 60 °C for four hours

Sulphated ash Not more than 0,1 % expressed on dry weight basis

Reducing sugars Not more than 0,2 % expressed as glucose on dry

weight basis

Other polyhydric alcohols Not more than 1 % expressed on dry weight basis

Nickel Not more than 2 mg/kg expressed on dry weight basis

Arsenic Not more than 3 mg/kg expressed on dry weight basis

Lead Not more than 1 mg/kg expressed on dry weight basis

Heavy metals Not more than 10 mg/kg expressed as Pb on dry weight

basis

Chlorides Not more than 100 mg/kg expressed on dry weight basis

Sulphates Not more than 200 mg/kg expressed on dry weight basis

▼<u>M3</u>

E 950 — ACESULFAME K

Synonyms Acesulfame potassium, potassium salt of 3,4-dihydro-6-

methyl-1,2,3-oxathiazin-4-one, 2,2-dioxide

Definition

Chemical name 6-methyl-1,2,3-oxathiazin-4(3H)-one-2,2-dioxide potas-

sium salt

Einecs 259-715-3

Chemical formula C₄H₄KNO₄S

Molecular weight 201,24

Assay Content not less than 99 % of C₄H₄KNO₄S on the anhy-

drous basis

Description Odourless, white, crystalline powder. Approximately

200 times as sweet as sucrose

Identification

A. Solubility Very soluble in water, very slightly soluble in ethanol

B. Ultra violet absorption Maximum 227 \pm 2 nm for a solution of 10 mg in

1 000 ml of water

C. Positive test for potassium

Passes test (test the residue obtained by igniting 2 g of

the sample)

D. Precipitation test

Add a few drops of a 10 % solution of sodium cobaltnitrite to a solution of 0,2 g of the sample in 2 ml of

acetic acid and 2 ml of water. A yellow precipitate is

produced

Purity

Loss on drying Not more than 1 % (105 °C, two hours)

Organic impurities Passes test for 20 mg/kg of UV active components

Fluoride Not more than 3 mg/kg
Lead Not more than 1 mg/kg

E 951 — ASPARTAME

Synonyms Aspartyl phenylalanine methyl ester

Definition

Chemical name N-L-α-(Aspartyl-L-phenylalanine-1-methyl ester, 3-

amino-N-(α-carbomethoxy-phenethyl)-succinamic acid-

N-methyl ester

Einecs 245-261-3 E number E 951 Chemical formula $C_{14}H_{18}N_2O_5$ Relative molecular mass 294.31

Assay Not less than 98 % and not more than 102 % of

 $C_{14}H_{18}N_2O_5$ on the anhydrous basis

Description White, odourless, crystalline powder having a sweet

taste. Approximately 200 times as sweet as sucrose

Identification

Solubility Slightly soluble in water and in ethanol

Purity

Loss on drying Not more than 4,5 % (105 °C, four hours)

Sulphated ash Not more than 0,2 % expressed on dry weight basis

pH Between 4,5 and 6,0 (1 in 125 solution)

Transmittance of a 1 % solution in 2N hydrochloric

acid, determined in a 1-cm cell at 430 nm with a suitable spectrophotometer, using 2N hydrochloric acid as a reference, is not less than 0,95, equivalent to an absorbance of not more than approximately 0,022

absorbance of not more than approximately 0,022

Specific rotation $(\alpha)_D^{20}$: + 14,5 to + 16,5°

Determine in a 4 in 100/15 N formic acid solution within 30 minutes after preparation of the sample solu-

tion

Arsenic Not more than 3 mg/kg expressed on dry weight basis

Lead Not more than 1 mg/kg expressed on dry weight basis

Heavy metals Not more than 10 mg/kg expressed as Pb on dry weight

basis

5-Benzyl-3,6-dioxo-2-piperazineacetic Not mo

acid

Not more than 1,5 % expressed on dry weight basis

E 952 — CYCLAMIC ACID AND ITS Na AND Ca SALTS

(I) CYCLAMIC ACID

Synonyms Cyclohexylsulphamic acid, cyclamate

Definition

Chemical name Cyclohexanesulphamic acid, cyclohexylaminosulphonic

acid

Einecs 202-898-1 E number E 952 Chemical formula $C_6H_{13}NO_3S$

Relative molecular mass 179,24

Assay Cyclohexylsulphamic acid contains not less than 98 %

and not more than the equivalent of 102 % of C_cH_1 , NO,S, calculated on the anhydrous basis

Description A practically colourless, white crystalline powder with a

sweet-sour taste. Approximately 40 times as sweet as

sucrose

Identification

A. Solubility

Soluble in water and in ethanol

B. Precipitation test

Acidify a 2 % solution with hydrochloric acid, add 1 ml of an approximately molar solution of barium chloride in water and filter if any haze or precipitate forms. To the clear solution add 1 ml of a 10 % solution of sodium nitrite. A white precipitate forms.

Purity

Loss on drying

Not more than 1 % (105 °C, one hour)

Selenium

Not more than 30 mg/kg expressed as selenium on dry

weight basis

Lead

Arsenic

Not more than 1 mg/kg expressed on dry weight basis Not more than 10 mg/kg expressed as Pb on dry weight

1. . . .

Heavy metals

Not more than 3 mg/kg expressed on dry weight basis

Cyclohexylamine
Dicyclohexylamine

Not more than 10 mg/kg expressed on dry weight basis Not more than 1 mg/kg expressed on dry weight basis

Aniline

Not more than 1 mg/kg expressed on dry weight basis

(II) SODIUM CYCLAMATE

Synonyms

Cyclamate, sodium salt of cyclamic acid

Definition

Chemical name

Sodium cyclohexanesulphamate, sodium cyclohexylsul-

phamate

Einecs E number 205-348-9 E 952

Chemical formula

C₆H₁₂NNaO₂S and the dihydrate form

 $C_6^0H_{12}^{12}NNaO_3^3S\cdot 2H_2O$

Relative molecular mass

201,22 calculated on the anhydrous form 237,22 calculated on the hydrated form

Assay

Not les sthan 98 % and not more than 102 % on the

dried basis

Dihydrate form: not less than 84 % on the dried basis

Description

White, odourless crystals or crystalline powder

Approximately 30 times as sweet as sucrose

Identification

Solubility

Soluble in water, practically insoluble in ethanol

Purity

Loss on drying

Not more than 1 % (105 °C, one hour)

Not more than 15,2 % (105 °C, two hours) for the dihy-

drate form

Selenium

Not more than 30 mg/kg expressed as selenium on dry

weight basis

Arsenic

Not more than 3 mg/kg expressed on dry weight basis Not more than 1 mg/kg expressed on dry weight basis

Lead

Heavy metals

Not more than 10 mg/kg expressed as Pb on dry weight

basis

Cyclohexylamine

Not more than 10 mg/kg expressed on dry weight basis

Dicyclohexylamine

Not more than 1 mg/kg expressed on dry weight basis

Aniline

Not more than 1 mg/kg expressed on dry weight basis

(III) CALCIUM CYCLAMATE

Cyclamate, calcium salt of cyclamic acid **Synonyms**

Definition

Chemical name Calcium cyclohexanesulphamate, calcium cyclohexylsul-

phamate

Einecs 205-349-4 E 952 E number

Chemical formula C₁₂H₂₄CaN₂O₆S₂·2H₂O

Relative molecular mass

Not less than 98 % and not more than 10 % on the dried Assay

Description White, colourless crystals or crystaline powder.

Approximately 30 times as sweet as sucrose

Identification

Solubility Soluble in water, sparingly soluble in ethanol

Purity

Not more than 1 % (105 °C, one hour) Loss on drying

Not more than 8,5 % (140 °C, four hours) for the dihy-

drate form

Selenium Not more than 30 mg/kg expressed as selenium on dry

weight basis

Arsenic Not more than 3 mg/kg expressed on dry weight basis Lead Not more than 1 mg/kg expressed on dry weight basis Not more than 10 mg/kg expressed as Pb on dry weight Heavy metals

Cyclohexylamine Not more than 10 mg/kg expressed on dry weight basis Not more than 1 mg/kg expressed on dry weight basis Dicyclohexylamine Aniline Not more than 1 mg/kg expressed on dry weight basis

E 954 — SACCHARIN AND ITS Na. K AND Ca SALTS

(I) SACCHARIN

Definition

Chemical name 3-oxo-2,3dihydrobenzo(d)isothiazol-1,1-dioxide

Einecs 201-321-0 E number E 954 Chemical formula C₇H₅NO₃S Relative molecular mass 183,18

Not less than 99 % and not more than 101,0 % of Assay

C₇H₅NO₃S on the anhydrous basis

White crystals or a white crystalline powder, odourless or with a faint, aromatic odour having a sweet taste even in very dilute solutions. Approximately between

300 and 500 times as sweet as sucrose

Identification

Description

Solubility Slightly soluble in water, soluble in basic solutions,

sparingly soluble in ethanol

Purity

Loss on drying Not more than 1 % (105 °C, two hours)

Melting range 226 to 230 °C Arsenic Not more than 3 mg/kg expressed on dry weight basis

Lead Not more than 1 mg/kg expressed on dry weight basis

Not more than 10 mg/kg expressed as Pb on dry weight

Not more than 30 mg/kg expressed on dry weight basis

Sulphated ash Not more than 0,2 % expressed on dry weight basis

Benzoic and salicylic acid To 10 ml of a 1 in 20 solution, previously acidified with five drops of acetic acid, add three drops of an approximately molar solution of ferric chloride in water. No

precipitate or violet colour appears

o-Toluenesulphonamide Not more than 10 mg/kg expressed on dry weight basis

p-Toluenesulphonamide Not more than 10 mg/kg expressed on dry weight basis

Benzoic acid p-sulfonamide Not more than 25 mg/kg expressed on dry weight basis

Readily carbonizable substances Absent

(II) SODIUM SACCHARIN

Selenium

Heavy metals

Synonyms Saccharin, sodium salt of saccharin

Definition

Chemical name Sodium o-benzosulphimide, sodium salt of 2,3-dihydro-3-oxobenzisosulfonazole, oxobenzisosulfonazole, 1,2-

benzisothiazolin-3-one-1, 1-dioxide sodium salt dihy-

drate

Einecs 204-886-1

E number E 954

Chemical formula C₇H₄NNaO₃S·2H₂O

Relative molecular mass 241,19

Assay Not less than 99 % and not more than 101 % of

C₇H₄NNaO₃S on the anhydrous basis

Description White crystals or a white crystalline efflorescent

> powder, odourless or with a faint, odour, hvaing an intensely sweet taste, even in very dilute solutions. Approximately between 300 and 500 times as sweet as

sucrose in dilute solutions

Identification

Solubility Freely soluble in water, sparingly soluble in ethanol

Purity

Loss on drying Not more than 15 % (120 °C, four hours)

Arsenic Not more than 3 mg/kg expressed on dry weight basis Selenium Not more than 30 mg/kg expressed on dry weight basis

Lead Not more than 1 mg/kg expressed on dry weight basis

Heavy metals Not more than 10 mg/kg expressed as Pb on dry weight

basis

Benzoic and salicylic acid To 10 ml of a 1 in 20 solution, previously acidified with five drops of acetic acid, add three drops of an approxi-

mately molar solution of ferric chloride in water. No precipitate or violet colour appears

o-Toluenesulphonamide Not more than 10 mg/kg expressed on dry weight basis p-Toluenesulphonamide Not more than 10 mg/kg expressed on dry weight basis Benzoic acid p-sulfonamide Not more than 25 mg/kg expressed on dry weight basis

Readily carbonizable substances Absent

(III) CALCIUM SACCHARIN

Synonyms Saccharin, calcium salt of saccharin Definition

Chemical name

Calcium o-benzosulphimide, calcium salt of 2,3-dihydro-3-oxobenzisosulphonazole, 1,2-benzisothiazolin-

3-one-1,1-dioxide calcium salt hydrate (2:7)

Einecs 229-349-0

E number E 954

Chemical formula $C_{14}H_8CaN_2O_6S_2\cdot 3\frac{1}{2}H_2O$

Relative molecular mass 467,48

Assay Not less than 95 % of $C_{14}H_8CaN_2O_6S_2$ on the anhydrous

basis

Description White crystals or a white crystalline powder, odourless

or with a faint odour, having an intensely sweet taste, even in very dilute solutions. Approximately between 300 and 500 times as sweet as sucrose in dilute solutions

Identification

Solubility Freely soluble in water, soluble in ethanol

Purity

Loss on drying Not more than 13,5 % (120 °C, four hours)

Arsenic Not more than 3 mg/kg expressed on dry weight basis

Selenium Not more than 30 mg/kg expressed on dry weight basis

Lead Not more than 1 mg/kg expressed on dry weight basis

Heavy metals Not more than 10 mg/kg expressed as Pb on dry weight

basis

Benzoic and salicylic acid

To 10 ml of a 1 in 20 solution, previously acidified with five drops of acetic acid, add three drops of an approxi-

mately molar solution of ferric chloride in water. No

precipitate or violet colour appears

o-Toluenesulphonamide Not more than 10 mg/kg expressed on dry weight basis

p-Toluenesulphonamide Not more than 10 mg/kg expressed on dry weight basis

Benzoic acid p-sulfonamide Not more than 25 mg/kg expressed on dry weight basis

Readily carbonizable substances Absent

(IV) POTASSIUM SACCHARIN

Synonyms Saccharin, potassium salt of saccharin

Definition

Chemical name Potassium o-benzosulphimide, potassium salt of 2,3-

dihydro-3-oxobenzisosulphonazole, potassium salt of 1,2-benzisothiazolin-3-one-1,1-dioxide monohydrate

Einecs

E number E 954

Chemical formula C₂H₄KNO₃S·H₂O

Relative molecular mass 239,77

Assay Not less than 99 % and not more than 101 % of

 $\mathrm{C_7H_4KNO_3S}$ on the anhydrous basis

DescriptionWhite crystals or a white crystalline powder, odourless or with a faint odour, having an intensely sweet tests

or with a faint odour, having an intensely sweet taste, even in very dilute solutions. Approximately between

300 and 500 times as sweet as sucrose

Identification

Solubility Freely soluble in water, sparingly soluble in ethanol

Purity

Loss on drying Not more than 8 % (120 °C, four hours)

Arsenic Not more than 3 mg/kg expressed on dry weight basis

Selenium Not more than 30 mg/kg expressed on dry weight basis

Lead Not more than 1 mg/kg expressed on dry weight basis

Heavy metals Not more than 10 mg/kg expressed as Pb on dry weight

basis

Benzoic and salicylic acid To 10 ml of a 1 in 20 solution, previously acidified with

five drops of acetic acid, add three drops of an approximately molar solution of ferric chloride in water. No

precipitate or violet colour appears

o-Toluenesulphonamide Not more than 10 mg/kg expressed on dry weight basis

p-Toluenesulphonamide Not more than 10 mg/kg expressed on dry weight basis

Benzoic acid p-sulfonamide Not more than 25 mg/kg expressed on dry weight basis

Readily carbonizable substances Absent

▼<u>M4</u>

E 955 — SUCRALOSE

Synonyms 4,1',6'-trichlorogalactosucrose

Definition

Chemical name 1,6-Dichloro-1,6-dideoxy-b-D-fructofuranosyl-4-chloro-

4-deoxy-a-D-galactopyranoside

Einecs 259-952-2

Chemical formula $C_{12}H_{19}Cl_3O_8$

Molecular weight 397,64

Assay Content not less than 98 % and not more than 102 %

C₁₂H₁₉Cl₃O₈ calculated on an anhydrous basis.

Description White to off-white, practically odourless crystalline

powder.

Identification

A. pH of a 10 % solution Not less than 5,0 and not more than 7,0

B. Solubility Freely soluble in water, methanol and ethanol

Slightly soluble in ethyl acetate

C. Infrared absorption

The infrared spectrum of a potassium bromide disper-

sion of the sample exhibits relative maxima at similar wave numbers as those shown in the reference spectrum

obtained using a sucralose reference standard.

D. Thin-layer chromatography

The main spot in the test solution has the same Rf value as that of the main spot of standard solution A referred

to in the test for other chlorinated disaccharides. This standard solution is obtained by dissolving 1,0 g of sucralose reference standard in 10 ml of methanol.

E. Specific rotation $[\alpha]^{20}D$: + 84,0° to + 87,5° calculated on the anhydrous

Ja818

(10 % w/v solution)

Purity

Water Not more than 2,0 % (Karl Fischer method)

Sulphated ash Not more than 0,7 %

Lead Not more than 1 mg/kg

Other chlorinated disaccharides

Not more than 0,5 %

Chlorinated monosaccharides

Not more than 0,1 %

Triphenylphosphine oxide Not more than 150 mg/kg

▼M4

Methanol Not more than 0,1 %

₹B

E 957 — THAUMATIN

Synonyms

Definition

Chemical name Thaumatin is obtained by aqueous extraction (pH 2,5 to

4) of the arils of the fruit of the natural strain of *Thau-matococcus daniellii* (Benth) and consists essentially of the proteins thaumatin I and thaumatin II together with minor amounts of plant constituents derived from the

source material

Einecs 258-822-2 E number E 957

Chemical formula Polypeptide of 207 aminoacids

Relative molecular mass Thaumatin I 22209

Thaumatin II 22293

Assay Not less than 16 % nitrogen on the dried basis equiva-

lent to not less than 94 % proteins (N × 5,8)

Description Odourless, cream-coloured powder with an intensely

sweet taste. Approximately 2 000 to 3 000 times as

sweet as sucrose

Identification

Solubility Very soluble in water, insoluble in acetone

Purity

Loss on drying

Not more than 9 % (105 °C to constant weight)

Not more than 3 % expressed on dry weight basis

Sulphated ash

Not more than 2 % expressed on dry weight basis

Not more than 100 mg/kg expressed on dry weight basis

Arsenic Not more than 3 mg/kg expressed on dry weight basis

Lead 3 mg/kg expressed on dry weight basis

Microbiological criteria Total aerobic microbial count: Max 1 000/g E. Coli:

absent in 1 g

E 959 — NEOHESPERIDINE DIHYDROCHALCONE

Synonyms Neohesperidin dihydrochalcone, NHDC, hesperetin dihydrochalcone-4'-β-neohesperidoside, neohesperidin

DC

Definition

Chemical name 2-O-α-L-rhamnopyranosyl-4'-β-D-glucopyranosyl

hesperetin dihydrochalcone; obtained by catalytic hydro-

genation of neohesperidin

Einecs 243-978-6
E number E 959
Chemical formula $C_{28}H_{36}O_{15}$ Relative molecular mass 612,6

Assay Content not less than 96 % on the dried basis

Description Off white, odourless, crystalline powder having a char-

acteristic, intensive sweet taste. Approximately between

1 000 and 1 800 times as sweet as sucrose

Identification

A. Solubility Freely soluble in hot water, very slightly soluble in cold

water, practically insoluble in ether and benzene

B. Ultraviolet absorption maximum 282 to 283 nm for a solution of 2 mg in 100 ml

methanol

C. Neu's test

Dissolve about 10 mg of neohesperidine DC in 1 ml methanol, add 1 ml of a 1 % 2-aminoethyl diphenyl

borate methanolic solution. A bright yellow colour is

produced

Purity

Loss on drying Not more than 11 % (105 °C, three huors)

Sulphated ash Not more than 0,2 % expressed on dry weight basis

Arsenic Not more than 3 mg/kg expressed on dry weight basis

Lead Not more than 2 mg/kg expressed on dry weight basis

Heavy metals Not more than 10 mg/kg expressed as Pb on dry weight

basis

▼<u>M4</u>

E 962 — SALT OF ASPARTAME-ACESULFAME

Synonyms Aspartame-acesulfame

Aspartame-acesulfame salt

DefinitionThe salt is prepared by heating an approximately 2:1 ratio (w/w) of aspartame and acesulfame K in solution at acidic pH and allowing crystallisation to occur. The

potassium and moisture are eliminated. The product is

more stable than aspartame alone.

Chemical name 6-methyl-1,2,3-oxathiazine-4(3H)-one-2,2-dioxide salt of

L-phenylalanyl-2-methyl-L-a-aspartic acid

Chemical formula $C_{18}H_{23}O_9N_3S$ Molecular weight 457,46

Assay 63,0 % to 66,0 % aspartame (dry basis) and 34,0 % to

37,0 % acesulfame (acid form on a dry basis)

Description A white, odourless, crystalline powder.

Identification

A. Solubility Sparingly soluble in water; slightly soluble in ethanol.

B. Transmittance The transmittance of a 1 % solution in water determined in a 1 cm cell at 430 nm with a suitable spectrophotometer using water as a reference, is not less than 0,95,

ometer using water as a reference, is not less than 0,95, equivalent to an absorbance of not more than approximately 0,022.

matery 0,02

C. Specific rotation $[\alpha]^{20}D: +14,5^{\circ} \text{ to } +16,5^{\circ}$

Determine at a concentration of 6,2 g in 100 ml formic acid (15N) within 30 min of preparation of the solution. Divide the calculated specific rotation by 0,646 to correct for the aspartame content of the salt of aspar-

tame-acesulfame.

Purity

Loss on drying Not more than 0,5 % (105 °C, 4 h)

5-Benzyl-3,6-dioxo-2-piperazineacetic Not more than 0,5 %

acid

Lead Not more than 1 mg/kg