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COUNCIL DIRECTIVE

of 25 July 1978

laying down specific criteria of purity for emulsifiers, stabilizers, thickeners and gelling agents for

use in foodstuffs

(78/663/EEC)

(OJ L 223, 14.8.1978, p. 7)

Amended by:

►<u>B</u>

		Official Journal		
		No	page	date
► <u>M1</u>	Council Directive 82/504/EEC of 12 July 1982	L 230	35	5.8.1982
► <u>M2</u>	Commission Directive 90/612/EEC of 26 October 1990	L 326	58	24.11.1990
► <u>M3</u>	Commission Directive 92/4/EEC of 10 February 1992	L 55	96	29.2.1992

Corrected by:

- ►C1 Corrigendum, OJ L 296, 21.10.1978, p. 50 (78/663)
- ▶ <u>C2</u> Corrigendum, OJ L 91, 10.4.1979, p. 7 (78/663)

COUNCIL DIRECTIVE

of 25 July 1978

laying down specific criteria of purity for emulsifiers, stabilizers, thickeners and gelling agents for use in foodstuffs (78/663/EEC)

THE COUNCIL OF THE EUROPEAN COMMUNITIES,

Having regard to the Treaty establishing the European Economic Community,

Having regard to Council Directive 74/329/EEC of 18 June 1974 on the approximation of the laws of the Member States relating to emulsifiers, stabilizers, thickeners and gelling agents for use in foodstuffs (¹), as last amended by Directive 78/612/EEC (²), and in particular Article 7 (1) thereof,

Having regard to the proposal from the Commission,

Whereas under Article 6 of Directive 74/329/EEC emulsifiers, stabilizers, thickeners and gelling agents must satisfy specific criteria of purity established in accordance with Article 7 (1) of that Directive,

HAS ADOPTED THIS DIRECTIVE:

Article 1

The specific criteria of purity referred to in Article 6 (1) (b) of Directive 74/329/EEC are given in the Annex to this Directive.

▼ M1

Article 2

As regards the substances referred to in the Annex under E 477, Member States may, until 31 December 1984, authorize for use in foodstuffs a product containing not more than 4 % dimer and trimer of propane-1,2-diol.

▼B

Article 3

Member States shall bring into force the laws, regulations and administrative provisions necessary to comply with this Directive not later than 18 months after the notification of this Directive. They shall forthwith inform the Commission thereof.

Article 4

This Directive is addressed to the Member States.

^{(&}lt;sup>1</sup>) OJ No L 189, 12. 7. 1974, p. 1. (²) OJ No L 197, 22. 7. 1978, p. 22.

ANNEX

SPECIFIC CRITERIA OF PURITY FOR EMULSIFIERS, STABILIZERS, THICKENERS AND GELLING AGENTS FOR USE IN FOODSTUFFS

General observations

- (a) Where interpretation of the criteria set out below requires the definition of certain technical details, reference should be made to the methods of analysis established pursuant to Article 7 (2) of Directive 74/329/EEC.
- (b) Unless otherwise stated, the quantities and percentages shall be calculated in terms of weight of the product as such.
- (c) The specific criteria of purity applicable to substances E 322, E 339 (i), (ii) and (iii), E 340 (i), (ii) and (iii) and E 341 (i) and (ii) are laid down by Council Directive 78/664/EEC of 25 July 1978 laying down specific criteria of purity for antioxidants which may be used in foodstuffs intended for human consumption (¹). The regime applicable to hydrolysed lecithins is set out in the same Directive.

E 341 — (iii) Tricalcium orthophosphate

Chemical description	— tricalcium diorthophosphate; Ca_3 (PO ₄) ₂ , — hydroxyapatite; Ca_5 (PO ₄) ₃ OH.
Appearance	Impalpable white powder.
Content	Not less than 90 % expressed as $Ca_3(PO_4)_2$ after calcination at 800 \pm 25 °C to constant weight.
Volatile matter	Not more than 10 % determined by calcination at 800 \pm 25 °C to constant weight.
Fluoride	Not more than 50 mg/kg expressed as fluorine.
E 400 — Alginic acid	
Chemical description	Linear glycuronoglycan consisting mainly of beta (1-4) linked D-mannuronic and alpha (1-4) linked L-guluronic acid units in pyranose ring form. Hydrophilic colloidal carbohydrate extracted by the use of dilute alkali from various species of brown seaweeds.
Description	Nearly odourless, tasteless, white to yellowish fibrous powder.
Content	Yields, on a volatile matter-free basis, not less than 20 % and not more than 23 % of carbon dioxide corresponding to not less than 91.0 % and not more than 104.5 % of alginic acid (equivalent weight 200).
Ash	Not more than 4 % on a volatile matter-free basis determined at 600 $^{\circ}$ C after drying at 105 $^{\circ}$ C for four hours.
Volatile matter	Not more than 15 % determined by drying at 105 °C for four

hours.

Acid-insoluble ash (insoluble in approximately 3 N hydrochloric acid)

▼<u>M1</u>

▼B

► M1 Not more than 2 %. <

▼B

▼B		
_	E 401 — Sodium alginate	
	Chemical name	Sodium salt of alginic acid.
	Description	Nearly odourless, tasteless white to yellowish fibrous or gran- ular powder.
	Content	Yields, on a volatile matter-free basis, not less than 18 % and not more than 21 % of carbon dioxide corresponding to not less than 90.8 % and not more than 106.0 % of sodium alginate (equivalent weight 222).
	Ash	Not less than 18.0 % and not more than 27.0 % on a volatile matter-free basis determined at 600 $^{\circ}$ C after drying at 105 $^{\circ}$ C for four hours.
▼ <u>M1</u>		
▼ <u>B</u>	Volatile matter	Not more than 15 % determined by drying at 105 °C for four hours.
	Acid-insoluble ash (insoluble in approximately 3 N hydrochloric acid)	▶ <u>M1</u> Not more than 2 %. \triangleleft
	E 402 — Potassium alginate	
	Chemical name	Potassium salt of alginic acid.
	Description	Nearly odourless, tasteless white to yellowish fibrous or gran- ular powder.
	Content	Yields, on a volatile matter-free basis, not less than 16.5 % and not more than 19.5 % of carbon dioxide corresponding to not less than 89.2 % and not more than 105.5 % of potassium alginate (equivalent weight 238).
	Ash	Not less than 23 % and not more than 32 % on a volatile matter-free basis determined at 600 $^{\circ}$ C after drying at 105 $^{\circ}$ C for four hours.
▼ <u>M1</u>		

• <u> </u>	Volatile matter	Not more than 15 % determined by drying at 105 $^{\circ}\mathrm{C}$ for four hours.
	Acid-insoluble ash (insoluble in approximately 3 N hydrochloric acid)	▶ <u>M1</u> Not more than 2 %. \triangleleft
	E 403 — Ammonium alginate	
	Chemical name	Ammonium salt of alginic acid.
	Description	White to yellowish fibrous or granular powder.
	Content	Yields, on a volatile matter-free basis, not less than 18 % and not more than 21 % of carbon dioxide corresponding to not less than 88.7 % and not more than 103.6 % of ammonium alginate (equivalent weight 217).
	Ash	Not more than 4% on a volatile matter-free basis determined at 600 $^\circ$ C after drying at 105 $^\circ$ C for four hours
▼ <u>M1</u>		
▼ <u>₿</u>	Volatile matter	Not more than 15% determined by drying at 105 °C for four hours.

▼B

Acid-insoluble ash (insoluble in approximately 3 N hydrochloric acid)

E 404 — Calcium alginate

Chemical name	Calcium salt of alginic acid.
Description	Nearly odourless, tasteless white to yellowish fibrous or gran- ular powder.
Content	Yields, on a volatile matter-free basis, not less than 18 % and not more than 21 % of carbon dioxide corresponding to not less than 89.6 % and not more than 104.5 % of calcium alginate (equivalent weight 219).
Ash	Not less than 15 % and not more than 24 % on a volatile matter-free basis determined at 600 °C after drying at 105 °C for four hours.

▶ <u>M1</u> Not more than 2 %. \triangleleft

▼<u>M1</u>

▼

B	
	Vol

acid)

 Volatile matter
 Not more than 15 % determined by drying at 105 °C for four hours.

 Acid-insoluble ash (insoluble in approximately 3 N hydrochloric
 ▶ <u>M1</u> Not more than 2 %.

E 405 — Propane-1,2-diol alginate

Chemical description	Propane-1.2-diol ester of alginic acid; varies in composition according to its degree of esterification and the percentage of free and neutralized carboxyl groups in the molecule.
Description	Nearly odourless and tasteless, white to yellowish fibrous or granular powder.
Content	Yields, on a volatile matter-free basis, not less than 16 $\%$ and not more than 20 $\%$ of carbon dioxide.
Ash	Not more than 10 % on a volatile matter-free basis determined at 600 °C after drying at 105 °C for four hours.
Total propane-1,2-diol content	Not less than 15 % and not more than 36 %.
Free propane-1,2-diol content	Not more than 12 %.

▼<u>M1</u>

▼B

Volatile matter	Not more than 20 $\%$ determined by drying at 105 $^{\circ}\mathrm{C}$ for four hours.
Acid-insoluble ash (insoluble in approximately 3 N hydrochloric acid)	▶ <u>M1</u> Not more than 2 %. \triangleleft
E 406 — Agar	
Chemical description	A hydrophilic colloidal polygalactoside, about 90 % of the galactose molecules being of the D-form and 10 % of the L-form. On about every tenth D-galactopyranose unit one of the hydroxyl groups is esterified with sulphuric acid which is neutralized by calcium, magnesium, potassium or sodium. It is extracted from certain marine algae of the families <i>Gelidiaceae</i> and <i>Sphaerococcaceae</i> and related red algae of the class <i>Rhodophyceae</i> .
Description	It occurs as white to pale yellow powder, fibres or flakes and is either odourless, or has a slight characteristic odour and a muci- laginous taste.

V D		
_	Ash	Not more than 6.5 $\%$ determined at 550 °C on a volatile matter-free basis.
	Acid-insoluble ash (insoluble in approximately 3 N hydrochloric acid)	Not more than 0.5 $\%$ determined at 550 °C on a volatile matter-free basis.
	Gelatin and other proteins	Dissolve about 1 g of agar in 100 ml of boiling water and allow to cool to about 50 °C. To 5 ml of the solution add 5 ml of trinitrophenol solution (1 g of anhydrous trinitrophenol/100 ml of hot water). No turbidity appears within 10 minutes.
	Insoluble matter (in hot water)	Not more than 1 %.
	Volatile matter	Not more than 20 $\%$ determined by drying at 105 $^\circ \mathrm{C}$ for five hours.
	Starch and dextrins	Boil 100 mg of agar in 100 ml of water. Cool and add a few drops of iodine solution (14 g I_2 in a solution of 36 g KI in 100 ml H_2O , add three drops of HCL and dilute to 1 000 ml). No blue or red colour is produced.
	Water absorption	Place 5 g of agar in a 100 ml graduated cylinder, fill to the mark with water, mix and allow to stand at about 25 °C for 24 hours. Pour the contents of the cylinder through moistened glass wool, allowing the water to drain into a second 100 ml graduated cylinder. Not more than 75 ml of water is obtained.
	E 407 — Carrageenan	
	Chemical description	Carrageenan is obtained by aqueous extraction of seaweeds of <i>Gigartinaceae, Solieriaceae, Hypneaceae</i> and <i>Furcellariaceae,</i> families of the class <i>Rhodophyceae</i> (red seaweeds). No organic precipitants shall be used other than methanol, ethanol and isopropanol. Carrageenan consists chiefly of the potassium, sodium, magnesium and calcium salts of polysaccharide sulphate esters which, on hydrolysis, yield galactose and 3.6-anhydrogalactose. Carrageenan shall not be hydrolysed or otherwise chemically degraded.
	Description	Yellowish to colourless, coarse to fine powder which is practi- cally odourless and has a mucilaginous taste.
	Volatile matter	Not more than 12 % determined by drying at 105 $^{\circ}\mathrm{C}$ for four hours.
	Sulphate	Not less than 15 % and not more than 40 % on a volatile matter-free basis, expressed as SO_4 .
▼ <u>M2</u>	Acid-insoluble ash (insoluble in 10 % w/v hydrochloric acid)	Not more than 1 % dry matter.
	Acid-insoluble matter (insoluble in $1 \% v/v$ sulphuric acid)	Not more than 2 % dry matter.
▼ <u>B</u>	Ash	Not less than 15 % and not more than 40 % determined at 550 $^{\circ}\mathrm{C}$ on a volatile matter-free basis.
	Methanol, ethanol, isopropanol content	Not more than 1 % singly or in combination.
	Viscosity of a 1.5 % solution at 75 $^{\circ}C$	Not less than five centipoises.
	E 410 — Locust bean gum	
	Chemical description	Consists mainly of a high molecular weight hydrocolloidal poly- saccharide, composed of galactopyranose and mannopyranose units combined through glycosidic linkages, which may be described chemically as galactomannan.
	Description	Locust bean gum is the ground endosperm of the seeds of the carob tree, <i>Cerationia siliqua</i> (L.) Taub. (Fam. <i>Leguminosae</i>). It is a white to yellowish-white, nearly odourless powder.

Galactomannan content	Not less than 75 %.
Insoluble matter (in 0.4 N sulphuric acid)	Not more than 4 % after digestion for six hours.
Ash	Not more than 1.2 % determined at 800 °C.
Volatile matter	Not more than 14 $\%$ determined by drying to constant weight at 102 to 105 °C (three to five hours).
Protein (N \times 6.25)	Not more than 7 %.
E 412 — Guar gum	
Chemical description	Consists mainly of a high molecular weight hydrocolloidal poly- saccharide composed of galactopyranose and mannopyranose units combined through glycosidic linkages, which may be described chemically as galactomannan.
Description	Guar gum is the ground endosperm of the seeds of the guar plant, <i>Cyamopsis tetragonolobus</i> (L.) Taub. (Fam. <i>Leguminosae</i>). It is a white to yellowish-white, nearly odourless powder.
Galactomannan content	Not less than 75 %.
Insoluble matter (in 0.4 N sulphuric acid)	Not more than 4 % after digestion for six hours.
Ash	Not more than 1.5 % determined at 800 °C.
Volatile matter	Not more than 14 $\%$ determined by drying to constant weight at 102 to 105 °C (three to five hours).
Protein (N \times 6.25)	Not more than 7 %.
E 413 — Tragacanth	
Chemical description	Consists mainly of high molecular weight polysaccharides composed of galacto-arabans and acidic polysaccharides containing galacturonic acid groups.
Description	Tragacanth is a dried gummy exudate obtained from <i>Astragalus</i> gummifer Labillardiere, or other Asiatic species of <i>Astragalus</i> (Fam. Leguminosae). Unground tragacanth occurs as flattened, lamellated, frequently curved fragments or straight or spirally twisted linear pieces from 0.5 to 2.5 mm in thickness. It is white to pale yellow in colour. It is odourless and has an insipid, mucilaginous taste. <i>Powdered tragacanth</i> is white to yellowish-white in colour.
Viscosity of a 1 % solution at 25 °C	Not less than 250 centipoises.
Ash	Not more than 3.5 % determined at 550 °C.
Acid-insoluble ash (insoluble in approximately 3 N hydrochloric acid)	Not more than 0.5 % determined at 550 °C.
Karaya gum	Boil 1 g with 20 ml of water until a mucilage is formed. Add 5 ml of hydrochloric acid and again boil the mixture for five minutes. No permanent pink or red colour develops.
E 414 — Acacia	
Chemical description	Consists mainly of high molecular weight polysaccharides and their calcium, potassium and magnesium salts, which on hydrol- ysis yield arabinose, galactose, rhamnose and glucuronic acid. It is obtained as a dried gummy exudate from the stems and branches of <i>Acacia senegal</i> (L) Willd. or of related species of <i>Acacia</i> (Fam. <i>Leguminosae</i>).
Description	Unground acacia occurs as white, yellowish-white or pale pinkish spheroidal tears of varying sizes or in angular fragments. It is also available commercially in the form of white or yellowish-white flakes, granules or powder.

▼ <u>B</u>		
	Ash	Not more than 4 % determined at 550 °C.
	Acid-insoluble ash (insoluble in approximately 3 N hydrochloric acid)	Not more than 0.5 % determined at 550 °C.
	Insoluble matter (in approximately 3 N hydrochloric acid)	Not more than 1 %.
	Volatile matter	Not more than 15 % determined by drying at 105 $^{\circ}\mathrm{C}$ for five hours.
	Starch or dextrin	Boil a 1 in 50 solution of the gum and cool. To 5 ml add one drop of iodine solution (14 g of iodine in a solution of 36 g of potassium iodide in 100 ml of water, add three drops of hydro-chloric acid and dilute to 1 000 ml). No bluish or reddish colour is produced.
	Tannin	To 10 ml of a 1 in 50 solution add about 0.1 ml of ferric chloride solution (9 g FeCl_3 , $6\text{H}_2\text{O}$ made up to 100 ml with water). No blackish colouration or blackish precipitate is formed.
▼ <u>M1</u>	E 415 — Xanthan gum	
	Chemical description	Xanthan gum is a high molecular weight polysaccharide gum produced by pure-culture fermentation of a carbohydrate with <i>Xanthomonas campestris</i> , purified by recovery with ethanol or isopropanol, dried and milled. It contains D-glucose and D- mannose as the dominant hexose units, along with D-glucuronic acid and pyruvic acid, and is prepared as the sodium, potassium or calcium salt. Its solutions are neutral.
	Description	Cream-coloured powder.
	Content	Xanthan gum yields, on a volatile matter-free basis, not less than $4.2~\%$ and not more than $5.0~\%$ of carbon dioxide.
	Volatile matter	Not more than 15 % determined by drying at 105 °C for 2 1/2 hours.
	Ash	Not more than 16% on a volatile-matter-free basis determined at 600 $^{\circ}\mathrm{C}$ after drying at 105 $^{\circ}\mathrm{C}$ for four hours.
	Pyruvic acid	Not less than 1.5 %.
	Nitrogen	Not more than 1.5 %.
	Isopropanol	Not more than 750 mg/kg.
	Microbiological criteria	Viable cells of Xanthomonas campestris shall be absent.
▼ <u>B</u>	E 420 — (i) Sorbitol	
	Chemical name	D-sorbitol.
	Description	White hygroscopic crystalline powder, flakes or granules, having a sweet taste.
	Content	Sorbitol contains not less than 98 % of glycitols and not less than 91 % of D-sorbitol, on a dry-matter basis in each case. Glycitols are compounds with the structural formula CH ₂ OH (CHOH) _n CH ₂ OH where 'n' is an integer. That part of the product which is not D-sorbitol is composed mainly of mannitol, together with small quantities of other glycitols, where $n \leq 4$, and minor quantities of hydrogenated oligosaccharides.
	Water	Not more than 1 % (Karl Fischer).
	Reducing sugars	Not more than 0.3 % on a dryweight basis, expressed as dextrose.

Total sugars	Not more than 1 $\%$ on a dryweight basis, expressed as dextrose.
Sulphated ash	Not more than 0.1 % at 800 \pm 25 °C on a dryweight basis.
Sulphate	Not more than 0.01 % on a dryweight basis, expressed as $\mathrm{SO}_4\!.$
Chloride	Not more than 0.005 % on a dryweight basis, expressed as Cl.
Nickel	Not more than 2 mg/kg, expressed as Ni.
E 420 — (ii) Sorbitol syrup	
Description	Clear, colourless and sweet-tasting aqueous solution of sorbitol and hydrogenated oligosaccharides. That part of the product which is not D-sorbitol is composed mainly of hydrogenated oligosaccharides formed by the hydrogenation of glucose syrup used as raw material (in which case the syrup is non-crystal- lizing) or mannitol. Minor quantities of glycitols where $n \le 4$ may be present. Glycitols are compounds with the structural formula CH_2OH (CHOH) _n CH_2OH , where 'n' is an integer.
Content	Not less than 69 $\%$ total solids and not less than 50 $\%$ of D-sorbitol.
Reducing sugars	Not more than 0.3 $\%$ on a dryweight basis, expressed as dextrose.
Sulphated ash	Not more than 0.1 % on a dryweight basis (after ignition at 800 \pm 25 °C).
Sulphate	Not more than 0.01 $\%$ on a dryweight basis, expressed as $\mathrm{SO}_4\!.$
Chloride	Not more than 0.005 $\%$ on a dryweight basis, expressed as Cl.
Nickel	Not more than 2 mg/kg, expressed as Ni.
E 421 — Mannitol	
Chemical name	D-mannitol.
Description	White crystalline solid which is odourless and has a sweet taste.
Description Content	White crystalline solid which is odourless and has a sweet taste. Not less than 98 % of D-mannitol $(C_6H_{14}O_6)$ on a volatile matter-free basis.
-	Not less than 98 % of D-mannitol $(C_6H_{14}O_6)$ on a volatile
Content	Not less than 98 % of D-mannitol $(C_6H_{14}O_6)$ on a volatile matter-free basis.
Content Melting range Specific rotation [a] ²⁵	Not less than 98 % of D-mannitol ($C_6H_{14}O_6$) on a volatile matter-free basis. 165 to 169 °C.
Content Melting range Specific rotation $\begin{bmatrix} a \end{bmatrix} $ $ \begin{array}{c} 25 \\ D \end{array} $	Not less than 98 % of D-mannitol (C ₆ H ₁₄ O ₆) on a volatile matter-free basis. 165 to 169 °C. Not less than + 23.0° and not more than + 24.3°. Not more than 0.3 % determined by drying at 105 °C for four
Content Melting range Specific rotation [a] ²⁵ D Volatile matter	Not less than 98 % of D-mannitol (C ₆ H ₁₄ O ₆) on a volatile matter-free basis. 165 to 169 °C. Not less than + 23.0° and not more than + 24.3°. Not more than 0.3 % determined by drying at 105 °C for four hours.
Content Melting range Specific rotation [a] ²⁵ D Volatile matter Reducing sugars	Not less than 98 % of D-mannitol (C ₆ H ₁₄ O ₆) on a volatile matter-free basis. 165 to 169 °C. Not less than + 23.0° and not more than + 24.3°. Not more than 0.3 % determined by drying at 105 °C for four hours. Not more than 0.05 %, expressed as dextrose.
Content Melting range Specific rotation [a] ²⁵ D Volatile matter Reducing sugars Sulphate	Not less than 98 % of D-mannitol (C ₆ H ₁₄ O ₆) on a volatile matter-free basis. 165 to 169 °C. Not less than + 23.0° and not more than + 24.3°. Not more than 0.3 % determined by drying at 105 °C for four hours. Not more than 0.05 %, expressed as dextrose. Not more than 0.01 %, expressed as SO ₄ .
Content Melting range Specific rotation [a] 25 D Volatile matter Reducing sugars Sulphate Chloride	Not less than 98 % of D-mannitol (C ₆ H ₁₄ O ₆) on a volatile matter-free basis. 165 to 169 °C. Not less than + 23.0° and not more than + 24.3°. Not more than 0.3 % determined by drying at 105 °C for four hours. Not more than 0.05 %, expressed as dextrose. Not more than 0.01 %, expressed as SO ₄ . Not more than 0.007 %, expressed as Cl.
Content Melting range Specific rotation [a] 25 D Volatile matter Reducing sugars Sulphate Chloride Ash	Not less than 98 % of D-mannitol ($C_6H_{14}O_6$) on a volatile matter-free basis. 165 to 169 °C. Not less than + 23.0° and not more than + 24.3°. Not more than 0.3 % determined by drying at 105 °C for four hours. Not more than 0.05 %, expressed as dextrose. Not more than 0.01 %, expressed as SO ₄ . Not more than 0.007 %, expressed as Cl. Not more than 0.1 % determined at 800 ± 25 °C.
Content Melting range Specific rotation [a] ²⁵ D Volatile matter Reducing sugars Sulphate Chloride Ash Nickel	Not less than 98 % of D-mannitol ($C_6H_{14}O_6$) on a volatile matter-free basis. 165 to 169 °C. Not less than + 23.0° and not more than + 24.3°. Not more than 0.3 % determined by drying at 105 °C for four hours. Not more than 0.05 %, expressed as dextrose. Not more than 0.01 %, expressed as SO ₄ . Not more than 0.007 %, expressed as Cl. Not more than 0.1 % determined at 800 ± 25 °C.
Content Melting range Specific rotation [a] ²⁵ [b] Volatile matter Reducing sugars Sulphate Chloride Ash Nickel E 422 — Glycerol	 Not less than 98 % of D-mannitol (C₆H₁₄O₆) on a volatile matter-free basis. 165 to 169 °C. Not less than + 23.0° and not more than + 24.3°. Not more than 0.3 % determined by drying at 105 °C for four hours. Not more than 0.05 %, expressed as dextrose. Not more than 0.01 %, expressed as SO₄. Not more than 0.1 % determined at 800 ± 25 °C. Not more than 2 mg/kg, expressed as Ni.
Content Melting range Specific rotation [a] ²⁵ [b] D Volatile matter Reducing sugars Sulphate Chloride Ash Nickel E 422 — Glycerol Description	 Not less than 98 % of D-mannitol (C₆H₁₄O₆) on a volatile matter-free basis. 165 to 169 °C. Not less than + 23.0° and not more than + 24.3°. Not more than 0.3 % determined by drying at 105 °C for four hours. Not more than 0.05 %, expressed as dextrose. Not more than 0.01 %, expressed as SO₄. Not more than 0.107 %, expressed as Cl. Not more than 0.1 % determined at 800 ± 25 °C. Not more than 2 mg/kg, expressed as Ni. Clear, colourless hygroscopic syrupy liquid with a sweet taste accompanied by a sensation of heat to the tongue.
Content Melting range Specific rotation [a] ²⁵ [b] D Volatile matter Reducing sugars Sulphate Chloride Ash Nickel E 422 — Glycerol Description Content	 Not less than 98 % of D-mannitol (C₆H₁₄O₆) on a volatile matter-free basis. 165 to 169 °C. Not less than + 23.0° and not more than + 24.3°. Not more than 0.3 % determined by drying at 105 °C for four hours. Not more than 0.05 %, expressed as dextrose. Not more than 0.01 %, expressed as SO₄. Not more than 0.107 %, expressed as Cl. Not more than 0.1 % determined at 800 ± 25 °C. Not more than 2 mg/kg, expressed as Ni. Clear, colourless hygroscopic syrupy liquid with a sweet taste accompanied by a sensation of heat to the tongue. Not less than 98 % of glycerol (C₃H₈O₃).

V D		
_	Acrolein, glucose and ammonium compounds	Heat a mixture of 5 ml of glycerol and 5 ml of potassium hydroxide solution (1 in 10) at 60 °C for five minutes. It neither becomes yellow nor emits an odour of ammonia.
	Butanetriols	Not more than 0.2 %.
	Chlorinated compounds (expressed as Cl)	Not more than 0.003 %.
	Fatty acids and esters	Not more than 0.1 % calculated as butyric acid.
	Sulphated ash	Not more than 0.01 % determined at 800 \pm 25 °C.
	E 440 (a) — Pectin	
	Chemical description	Pectin consists mainly of the partial methyl esters of polygalac- turonic acid and their sodium, potassium, calcium and ammonium salts.
		Pectin is obtained by aqueous extraction of appropriate edible plant material, usually citrus fruits or apples. No organic precipi- tants shall be used other than methanol, ethanol and isopropanol.
	Description	White, light yellow, light grey or light brown powder.
	Galacturonic acid	Not less than 65 % calculated on an ash and volatile matter-free basis after washing with acid and alcohol.
▼ <u>C2</u>	Volatile matter	Not more than 12 % determined by drying at 105 $^{\circ}$ C for two hours.
▼ <u>B</u>	Acid-insoluble ash (insoluble in approximately 3 N hydrochloric acid)	Not more than 1 %.
	Free methanol, ethanol and isopropanol content	Not more than 1 %, singly or in combination, on a volatile matter-free basis.
	Sulphur dioxide residue	Not more than 50 mg/kg on a volatile matter-free basis.
	Nitrogen content	Not more than 0.5 % determined after washing with acid and alcohol (kjeldahl).
	E 440 (b) — Amidated pectin	
	Chemical description	Amidated pectin consists mainly of the partial methyl esters and amides of polygalacturonic acid and their ammonium, sodium, potassium and calcium salts. It is obtained by aqueous extraction of appropriate edible plant material, usually citrus fruits or apples and treatment with ammonia under alkaline conditions. No organic precipitants shall be used other than methanol, ethanol and isopropanol.
	Description	White, light yellow, light grey or light brown powder.
	Degree of amidation	Not more than 25 % of total carboxyl groups.
	Galacturonic acid	Not less than 65 % calculated on an ash and volatile matter-free basis determined after washing with acid and alcohol.
▼ <u>C2</u>	Volatile matter	Not more than 12 % determined by drying at 105 $^{\circ}\mathrm{C}$ for two hours.
▼ <u>B</u>	Acid-insoluble ash (insoluble in approximately 3 N hydrochloric acid)	Not more than 1 %.
	Free methanol, ethanol and isopropanol content	Not more than 1 %, singly or in combination, on a volatile matter-free basis.
	Sulphur dioxide residue	Not more than 50 mg/kg on a volatile matter-free basis.

Not more than 2.5 % after washing with acid and alcohol (kjeldahl). Nitrogen content

E 450 (a) — (i) Disodium dihydrogen diphosphate

Description	White powder or grains.	
Content	Not less than 95.0 % of $Na_2H_2P_2O_7$.	
Content in P_2O_5	Not less than 63.0 $\%$ and not more than 64.0 %.	
Volatile matter	Not more than 0.5 % determined by drying at 105 °C for four hours.	
pH of 1 % solution	Not less than 3.7 and not more than 4.4.	
Water insoluble matter	Not more than 0.6 %.	
Fluoride	Not more than 10 mg/kg expressed as fluorine.	
E 450 (a) — (ii) Trisodium dipho	osphate	
Description	White powder or grains. Occurs anhydrous or as a monohydrate.	
Content	Not less than 95.0 % of $Na_3HP_2O_7$ or of $Na_3HP_2O_7$, H_2O .	
Content in P_2O_5	Not less than 57.5 % and not more than 58.5 % for the anhydrous salt. Not less than 53.6 % and not more than 54.6 % for the monohydrate.	
pH of a 1 % solution	Not less than 6.7 and not more than 7.3.	
Volatile matter	Not more than 0.5 % determined by drying at 105 $^{\circ}\mathrm{C}$ for four hours.	
Water insoluble matter	Not more than 0.2 %.	
Fluoride	Not more than 10 mg/kg expressed as fluorine.	
E 450 (a) — (iii) Tetrasodium diphosphate		
E 450 (a) — (iii) Tetrasodium di	phosphate	
E 450 (a) — (iii) Tetrasodium dip Description	phosphate White, crystalline or granular powder. Occurs anhydrous or as a decahydrate.	
	White, crystalline or granular powder. Occurs anhydrous or as a	
Description	White, crystalline or granular powder. Occurs anhydrous or as a decahydrate.	
Description Content	White, crystalline or granular powder. Occurs anhydrous or as a decahydrate. Not less than 95.0 % of $Na_4P_2O_7$ or of $Na_4P_2O_7$, $10H_2O$. Not less than 52.5 % and not more than 54.0 % for the anhydrous salt. Not less than 31.5 % and not more than 32.5 % for the decahy-	
Description Content Content in P_2O_5	 White, crystalline or granular powder. Occurs anhydrous or as a decahydrate. Not less than 95.0 % of Na₄P₂O₇ or of Na₄P₂O₇, 10H₂O. Not less than 52.5 % and not more than 54.0 % for the anhydrous salt. Not less than 31.5 % and not more than 32.5 % for the decahydrate. Not more than 0.5 % for the anhydrous salt, not less than 38 % and not more than 42 % for the decahydrate, in both cases determined after drying at 105 °C for four hours, followed by 	
Description Content Content in P_2O_5 Loss on ignition	 White, crystalline or granular powder. Occurs anhydrous or as a decahydrate. Not less than 95.0 % of Na₄P₂O₇ or of Na₄P₂O₇, 10H₂O. Not less than 52.5 % and not more than 54.0 % for the anhydrous salt. Not less than 31.5 % and not more than 32.5 % for the decahydrate. Not more than 0.5 % for the anhydrous salt, not less than 38 % and not more than 42 % for the decahydrate, in both cases determined after drying at 105 °C for four hours, followed by ignition at 550 °C for 30 minutes. 	
Description Content Content in P_2O_5 Loss on ignition pH of a 1 % solution	 White, crystalline or granular powder. Occurs anhydrous or as a decahydrate. Not less than 95.0 % of Na₄P₂O₇ or of Na₄P₂O₇, 10H₂O. Not less than 52.5 % and not more than 54.0 % for the anhydrous salt. Not less than 31.5 % and not more than 32.5 % for the decahydrate. Not more than 0.5 % for the anhydrous salt, not less than 38 % and not more than 42 % for the decahydrate, in both cases determined after drying at 105 °C for four hours, followed by ignition at 550 °C for 30 minutes. Not less than 9.9 and not more than 10.7. 	
Description Content Content in P_2O_5 Loss on ignition pH of a 1 % solution Water insoluble matter	 White, crystalline or granular powder. Occurs anhydrous or as a decahydrate. Not less than 95.0 % of Na₄P₂O₇ or of Na₄P₂O₇, 10H₂O. Not less than 52.5 % and not more than 54.0 % for the anhydrous salt. Not less than 31.5 % and not more than 32.5 % for the decahydrate. Not more than 0.5 % for the anhydrous salt, not less than 38 % and not more than 42 % for the decahydrate, in both cases determined after drying at 105 °C for four hours, followed by ignition at 550 °C for 30 minutes. Not less than 9.9 and not more than 10.7. Not more than 0.2 %. Not more than 10 mg/kg expressed as fluorine. 	
Description Content Content in P_3O_5 Loss on ignition pH of a 1 % solution Water insoluble matter Fluoride	 White, crystalline or granular powder. Occurs anhydrous or as a decahydrate. Not less than 95.0 % of Na₄P₂O₇ or of Na₄P₂O₇, 10H₂O. Not less than 52.5 % and not more than 54.0 % for the anhydrous salt. Not less than 31.5 % and not more than 32.5 % for the decahydrate. Not more than 0.5 % for the anhydrous salt, not less than 38 % and not more than 42 % for the decahydrate, in both cases determined after drying at 105 °C for four hours, followed by ignition at 550 °C for 30 minutes. Not less than 9.9 and not more than 10.7. Not more than 0.2 %. Not more than 10 mg/kg expressed as fluorine. 	
Description Content Content in P_2O_5 Loss on ignition pH of a 1 % solution Water insoluble matter Fluoride E 450 (a) — (iv) Tetrapotassium	 White, crystalline or granular powder. Occurs anhydrous or as a decahydrate. Not less than 95.0 % of Na₄P₂O₇ or of Na₄P₂O₇, 10H₂O. Not less than 52.5 % and not more than 54.0 % for the anhydrous salt. Not less than 31.5 % and not more than 32.5 % for the decahydrate. Not more than 0.5 % for the anhydrous salt, not less than 38 % and not more than 42 % for the decahydrate, in both cases determined after drying at 105 °C for four hours, followed by ignition at 550 °C for 30 minutes. Not less than 9.9 and not more than 10.7. Not more than 10 mg/kg expressed as fluorine. 	
Description Content Content in P_2O_5 Loss on ignition pH of a 1 % solution Water insoluble matter Fluoride E 450 (a) — (iv) Tetrapotassium Description	 White, crystalline or granular powder. Occurs anhydrous or as a decahydrate. Not less than 95.0 % of Na₄P₂O₇ or of Na₄P₂O₇, 10H₂O. Not less than 52.5 % and not more than 54.0 % for the anhydrous salt. Not less than 31.5 % and not more than 32.5 % for the decahydrate. Not more than 0.5 % for the anhydrous salt, not less than 38 % and not more than 42 % for the decahydrate, in both cases determined after drying at 105 °C for four hours, followed by ignition at 550 °C for 30 minutes. Not less than 9.9 and not more than 10.7. Not more than 0.2 %. Not more than 10 mg/kg expressed as fluorine. diphosphate Colourless crystals or white, very hygroscopic powder. 	

pH of a 1 % solution	Not less than 10.0 and not more than 10.7.
Water insoluble matter	Not more than 0.2 %.
Fluoride (expressed as F)	Not more than 10 mg/kg.
E 450 (b) — (i) Pentasodium tri	phosphate
Description	White, slightly hygroscopic granules or powder. Occurs anhy- drous or as a hexahydrate.
Content	Not less than 85.0 % of $Na_5P_3O_{10}$ or of $Na_5P_3O_{10}$ · $6H_2O$, the remainder being principally other sodium phosphates(E 450).
Content in P_2O_5	Not less than 56.0 % and not more than 58.0 % for the anhydrous salt. Not less than 43.0 % and not more than 45.0 % for the hexahydrate.
Loss on ignition	Not more than 0.5 % for the anhydrous salt and not more than 23.5 % for the hexahydrate, in both cases determined after drying at 105 °C for four hours followed by ignition at 550 °C for 30 minutes.
pH of a 1 % solution	Not less than 9.3 and not more than 10.1.
Water insoluble matter	Not more than 0.2 %.
Fluoride (expressed as F)	Not more than 10 mg/kg.
E 450 (b) — (ii) Pentapotassium	triphosphate
Description	White, very hygroscopic powder.
Content	Not less than 85.0 % of $K_5 P_3 O_{10}$, the remainder being principally other potassium phosphates (E 450).
Content in P_2O_5	Not less than 46.5 % and not more than 48.0 %.
Loss on ignition	Not more than 0.5 % calculated on the P_2O_5 content after drying at 105 °C for four hours, followed by ignition at 550 °C for 30 minutes.
pH of a 1 % solution	Not less than 9.3 and not more than 10.1.
Water insoluble matter	Not more than 2 %.
Fluoride (expressed as F)	Not more than 10 mg/kg.
E 450 (c) — (i) Sodium polyphos	sphates
Chemical description	Heterogeneous mixtures of sodium salts of linear condensed polyphosphoric acids of general formula $H_{(n+2)} P_n O_{(3n+1)}$ where 'n' is not less than 2.
Description	Fine white powders or crystals or colourless glassy platelets.
Content in P_2O_5	Not less than 59.5 $\%$ and not more than 70.0 $\%,$ calculated on the ignited basis.
Loss on ignition	Not more than 0.5 % after drying at 105 °C for four hours followed by ignition at 550 °C for 30 minutes.
pH of a 1 % solution	Not less than 3.6 and not more than 9.0.
Water insoluble matter	Not more than 0.2 %.
Fluoride	Not more than 10 mg/kg expressed as fluorine.
Cyclic phosphates	Not more than 8 %.
E 450 (c) — (ii) Potassium polyp	phosphates

Chemical description

▼<u>B</u>

Heterogeneous mixtures of potassium salts of linear condensed polyphosphoric acids of general formula $H_{(n+2)}P_nO_{(3n+1)}$ where 'n' is not less than 2.

Description	Fine white powders or crystals or colourless glassy platelets.
Content in P_2O_5	Not less than 53.5 $\%$ and not more than 61.5 $\%,$ calculated on the ignited basis.
Loss on ignition	Not more than 2 % after drying at 105 °C for four hours followed by ignition at 550 °C for 30 minutes.
pH of a 1 % solution	Not more than 7.8 (¹).
Water insoluble matter	Not more than 0.2 % $(^1)$.
Fluoride	Not more than 10 mg/kg expressed as fluorine.
Cyclic phosphates	Not more than 8 %.
▶ <u>M1</u> E 460 — (i) ◀ Microcrys	talline cellulose
Chemical description	Microcrystalline cellulose is purified partially depolymerized cellulose prepared by acid hydrolysis of alpha-cellulose obtained directly from fibrous plant material. It has a molecular weight of about 36 000.
Description	A fine white or almost white odourless powder.
Volatile matter	Not more than 5 % determined by drying to constant weight at 105 °C.
рН	Shake about 5 g with 40 ml of carbon dioxide-free water for 20 minutes and centrifuge. The pH of the supernatant liquid is between 5.5 and 7.
Sulphated ash	Not more than 0.1 % determined at 800 \pm 25 °C.
Water soluble substances	Not more than 0.16 %.
Diethyl ether extractable matter	Not more than 200 mg/kg.
Chloride	Not more than 350 mg/kg expressed as Cl.
Sulphate	Not more than 600 mg/kg expressed as SO_4 .

▼<u>M1</u> E 460 — (ii) Powdered cellulose

Chemical description	Powdered cellulose is purified mechanically disintegrated cellulose prepared by processing alpha-cellulose obtained directly from fibrous plant material. It has a molecular weight of 1.6×10^5 or greater.
Description	A white, odourless powder.
Content	Not less than 92 % $(C_{12}H_{20}O_{10})_n$.
Volatile matter	Not more than 7 % determined by drying at 105 $^{\rm o}{\rm C}$ for three hours.
рН	Shake about 5 g with 40 ml of carbon-dioxide-free water for 20 minutes and centrifuge. The pH of the supernatant liquid is between 5.0 and 7.5 .
Sulphated ash	Not more than 0.3 % determined at 800 \pm 25 °C.
Water-soluble substances	Not more than 1 %.

▼<u>B</u>

E 461 — Methylcellulose

Chemical description	Methylcellulose is cellulose obtained directly from fibrous plant material and partially etherified with methyl groups.
Description	Slightly hygroscopic white or slightly yellowish or greyisho- dourless and tasteless, granular or fibrous powder.

Chemical formula	The polymers contain substituted anhydroglucose units with the following general formula:
	$C_6H_7O_2$ (OR ₁) (OR ₂) (OR ₃) where
	R_1, R_2, R_3 each may be
	— H, — CH ₄ , or
	$-CH_2 CH_2 OH.$
Molecular weight	From about 20 000 to 380 000.
Content of substituted groups	Not less than 25 % and not more than 33 % of methoxyl groups (-OCH ₃). Not more than 5 % of hydroxyethoxyl groups (-OCH ₂ CH ₂ OH).
Volatile matter	Not more than 10 $\%$ determined by drying to constant weight at 105 $^{\circ}\mathrm{C}.$
Sulphated ash	Not more than 1.5 % determined at 800 \pm 25 °C.
pH of a 1 % solution	Not less than 5 and not more than 8.
E 463 — Hydroxypropylcellulose	
Chemical description	Hydroxypropylcellulose is cellulose obtained directly from fibrous plant material and partially etherified with hydroxy- propyl groups.
Description	Slightly hygroscopic white or slightly yellowish or greyish odourless and tasteless, granular or fibrous powder.
Chemical formula	The polymers contain substituted anhydroglucose units with the following general formula:
	$C_6H_7O_2$ (OR ₁) (OR ₂) (OR ₃) where
	R_1 , R_2 , R_3 each may be any one of the following:
	— Н, — СН,СНОНСН,,
	- CH ₂ CHOHCH ₃ , CH ₂ CHOHCH ₃) CH ₃ ,
	$- CH_2CHO [CH_2CHO (CH_2CHOHCH_3) CH_3] CH_3.$
Molecular weight	From about 30 000 to 1 000 000.
Content of substituted groups	Not more than 80.5 % of hydroxypropoxyl groups $(-OCH_2CHOHCH_3)$ on a volatile matter-free basis, equivalent to not more than 4.6 hydroxypropyl groups per anhydroglucose unit.
pH of a 1 % solution	Not less than 5.0 and not more than 8.0.
Volatile matter	Not more than 10 $\%$ determined by drying to constant weight at 105 $^{\circ}\mathrm{C}.$
Sulphated ash	Not more than 0.5 % determined at 800 \pm 25 °C.
E 464 — Hydroxypropylmethylce	llulose
Chemical description	Hydroxypropylmethylcellulose is cellulose obtained directly from fibrous plant material and partially etherified with methyl groups and containing a small degree of hydroxypropyl substitu- tion.
Description	Slightly hygroscopic white or slightly yellowish or greyish odourless and tasteless, granular or fibrous powder.
Chemical formula	The polymers contain substituted anhydroglucose units with the following general formula:
	$C_6H_7O_2$ (OR ₁) (OR ₂) (OR ₃) where
	R_1 , R_2 and R_3 each may be any one of the following:
	- CH ₃ ,
	$-CH_2CHOHCH_3,$

	- CH ₂ CHO (CH ₂ CHOHCH ₃) CH ₃ , - CH ₂ CHO [CH ₂ CHO (CH ₂ CHOHCH ₃)CH ₃] CH ₃ .
Molecular weight	From about 13 000 to 200 000.
Content of substituted groups	Not less than 19 % and not more than 30 % of methoxyl groups (-OCH ₃) and not less than 3 % and not more than 12 % hydroxypropoxyl groups (-OCH ₂ CHOHCH ₃) on a volatile matter-free basis.
pH of a 1 % solution	Not less than 5.0 and not more than 8.0.
Volatile matter	Not more than 10 % determined by drying to constant weight at 105 °C.
Sulphated ash	Not more than 1.5 % for products with viscosities greater than 50 cP and not more than 3.0 % for products with viscosities of 50 cP or less, determined at 800 ± 25 °C.
E 465 — Ethylmethylcellulose	
Chemical description	Ethylmethylcellulose is cellulose obtained directly from fibrous plant material and partially etherified with methyl and ethyl groups.
Description	Slightly hygroscopic white or slightly yellowish or greyish odourless and tasteless, granular or fibrous powder.
Chemical formula	The polymers contain substituted anhydroglucose units with the following general formula:
	$C_6H_7O_2$ (OR ₁) (OR ₂) (OR ₃) where
	R_1 , R_2 and R_3 each may be any one of the following:
	$-CH_3,$ $-CH_2CH_3.$
Molecular weight	From about 30 000 to 40 000.
-	
Content of substituted groups	Not less than 14.5 % and not more than 19.0 % of ethoxyl groups (-OC ₂ H ₅) and not less than 3.5 % and not more than 6.5% of methoxyl groups (-OCH ₃) on a volatile matter-free basis.
C C	Not less than 14.5 % and not more than 19.0 % of ethoxyl groups $(-OC_2H_5)$ and not less than 3.5 % and not more than 6.5% of methoxyl groups $(-OCH_3)$ on a volatile matter-free
Content of substituted groups	Not less than 14.5 % and not more than 19.0 % of ethoxyl groups ($-OC_2H_5$) and not less than 3.5 % and not more than 6.5% of methoxyl groups ($-OCH_3$) on a volatile matter-free basis. Fibrous form: not more than 15 %. Powdered form: not more than 10 %. determined by drying to constant weight at 105 °C
Content of substituted groups Volatile matter	Not less than 14.5 % and not more than 19.0 % of ethoxyl groups ($-OC_2H_3$) and not less than 3.5 % and not more than 6.5% of methoxyl groups ($-OCH_3$) on a volatile matter-free basis. Fibrous form: not more than 15 %. Powdered form: not more than 10 %. determined by drying to constant weight at 105 °C in each case.
Content of substituted groups Volatile matter Sulphated ash	Not less than 14.5 % and not more than 19.0 % of ethoxyl groups ($-OC_2H_5$) and not less than 3.5 % and not more than 6.5% of methoxyl groups ($-OCH_3$) on a volatile matter-free basis. Fibrous form: not more than 15 %. Powdered form: not more than 10 %. determined by drying to constant weight at 105 °C in each case. Not more than 0.6 % determined at 800 ± 25 °C.
Content of substituted groups Volatile matter Sulphated ash pH of a 1 % solution	Not less than 14.5 % and not more than 19.0 % of ethoxyl groups ($-OC_2H_5$) and not less than 3.5 % and not more than 6.5% of methoxyl groups ($-OCH_3$) on a volatile matter-free basis. Fibrous form: not more than 15 %. Powdered form: not more than 10 %. determined by drying to constant weight at 105 °C in each case. Not more than 0.6 % determined at 800 ± 25 °C.
Content of substituted groups Volatile matter Sulphated ash pH of a 1 % solution E 466 — Carboxymethylcellulose	Not less than 14.5 % and not more than 19.0 % of ethoxyl groups ($-OC_2H_3$) and not less than 3.5 % and not more than 6.5% of methoxyl groups ($-OCH_3$) on a volatile matter-free basis. Fibrous form: not more than 15 %. Powdered form: not more than 10 %. determined by drying to constant weight at 105 °C in each case. Not more than 0.6 % determined at 800 ± 25 °C. Not less than 5 and not more than 8.
Content of substituted groups Volatile matter Sulphated ash pH of a 1 % solution E 466 — Carboxymethylcellulose Chemical description	Not less than 14.5 % and not more than 19.0 % of ethoxyl groups ($-OC_2H_5$) and not less than 3.5 % and not more than 6.5% of methoxyl groups ($-OCH_3$) on a volatile matter-free basis. Fibrous form: not more than 15 %. Powdered form: not more than 10 %. determined by drying to constant weight at 105 °C in each case. Not more than 0.6 % determined at 800 ± 25 °C. Not less than 5 and not more than 8. Carboxymethylcellulose is the partial sodium salt of a carboxymethyl ether of cellulose, the cellulose being obtained directly from fibrous plant material. Slightly hygroscopic white or slightly yellowish or greyish
Content of substituted groups Volatile matter Sulphated ash pH of a 1 % solution E 466 — Carboxymethylcellulose Chemical description Description	Not less than 14.5 % and not more than 19.0 % of ethoxyl groups ($-OC_2H_5$) and not less than 3.5 % and not more than 6.5% of methoxyl groups ($-OCH_3$) on a volatile matter-free basis. Fibrous form: not more than 15 %. Powdered form: not more than 10 %. determined by drying to constant weight at 105 °C in each case. Not more than 0.6 % determined at 800 ± 25 °C. Not less than 5 and not more than 8. Carboxymethylcellulose is the partial sodium salt of a carboxymethyl ether of cellulose, the cellulose being obtained directly from fibrous plant material. Slightly hygroscopic white or slightly yellowish or greyish odourless and tasteless, granular or fibrous powder.
Content of substituted groups Volatile matter Sulphated ash pH of a 1 % solution E 466 — Carboxymethylcellulose Chemical description Description	Not less than 14.5 % and not more than 19.0 % of ethoxyl groups ($-OC_2H_5$) and not less than 3.5 % and not more than 6.5% of methoxyl groups ($-OCH_3$) on a volatile matter-free basis. Fibrous form: not more than 15 %. Powdered form: not more than 10 %. determined by drying to constant weight at 105 °C in each case. Not more than 0.6 % determined at 800 ± 25 °C. Not less than 5 and not more than 8. Carboxymethylcellulose is the partial sodium salt of a carboxy-methyl ether of cellulose, the cellulose being obtained directly from fibrous plant material. Slightly hygroscopic white or slightly yellowish or greyish odourless and tasteless, granular or fibrous powder. The polymers contain substituted anhydroglucose units with the following general formula: $C_6H_7O_2$ (OR_1) (OR_2) (OR_3) where R_1 , R_2 and R_3 each may be any one of the following:
Content of substituted groups Volatile matter Sulphated ash pH of a 1 % solution E 466 — Carboxymethylcellulose Chemical description Description	Not less than 14.5 % and not more than 19.0 % of ethoxyl groups ($-OC_2H_3$) and not less than 3.5 % and not more than 6.5% of methoxyl groups ($-OCH_3$) on a volatile matter-free basis. Fibrous form: not more than 15 %. Powdered form: not more than 10 %. determined by drying to constant weight at 105 °C in each case. Not more than 0.6 % determined at 800 ± 25 °C. Not less than 5 and not more than 8. Carboxymethylcellulose is the partial sodium salt of a carboxymethyl ether of cellulose, the cellulose being obtained directly from fibrous plant material. Slightly hygroscopic white or slightly yellowish or greyish odourless and tasteless, granular or fibrous powder. The polymers contain substituted anhydroglucose units with the following general formula: $C_6H_7O_2$ (OR_1) (OR_2) (OR_3) where
Content of substituted groups Volatile matter Sulphated ash pH of a 1 % solution E 466 — Carboxymethylcellulose Chemical description Description	Not less than 14.5 % and not more than 19.0 % of ethoxyl groups ($-OC_2H_3$) and not less than 3.5 % and not more than 6.5% of methoxyl groups ($-OCH_3$) on a volatile matter-free basis. Fibrous form: not more than 15 %. Powdered form: not more than 10 %. determined by drying to constant weight at 105 °C in each case. Not more than 0.6 % determined at 800 ± 25 °C. Not less than 5 and not more than 8. Carboxymethylcellulose is the partial sodium salt of a carboxy-methyl ether of cellulose, the cellulose being obtained directly from fibrous plant material. Slightly hygroscopic white or slightly yellowish or greyish odourless and tasteless, granular or fibrous powder. The polymers contain substituted anhydroglucose units with the following general formula: $C_6H_7O_2$ (OR_1) (OR_2) (OR_3) where R_1 , R_2 and R_3 each may be any one of the following: - H, $- CH_2COONa$,

Sodium chloride and sodium glycolate	Not more than 0.5 $\%$ total, and not more than 0.4 $\%$ of sodium glycolate.
Degree of substitution	Not less than 0.2 and not more than 1.0 carboxymethyl groups (-CH $_2$ COOH) per anhydroglucose unit.
Sodium	Not more than 9.7 % on a volatile matter-free basis.
Volatile matter	Not more than 12 $\%$ determined by drying to constant weight at 105 $^{\circ}\mathrm{C}.$
pH of a 1 % solution	Not less than 6 and not more than 8.5.
E 470 — Sodium, potassium and	calcium salts of fatty acids
Chemical description	Sodium, potassium and calcium salts of fatty acids occurring in food oils and fats, these salts being obtained either from edible fats or from distilled food fatty acids.
Description	White or creamy white light powders, flakes, or semi-solids.
Unsaponifiable matter	Not more than 2 %.
Free fatty acids	Not more than 3 % estimated as oleic acid.
Total glycerol (combined and free)	Not more than 10 %.
Free alkali	Not more than 0.1 % expressed as NaOH.
Matter insoluble in alcohol	Not more than 0.2 $\%$ (sodium and potassium salts only).
Volatile matter	Not more than 3 %.
Content of sodium, or potassium, or calcium	Sodium Not less than 9.0 % and not more than 14.0 % expressed as Na_2O . Potassium Not less than 13.0 % and not more than 21.5 % expressed as K_2O . Calcium Not less than 8.5 % and not more than 13.0 % expressed as CaO.
E 471 — Mono- and diglycerides	s of fatty acids
Chemical description	Mono- and diglycerides of fatty acids consist of mixtures of glycerol mono-, di- and tri-esters of fatty acids occurring in food fats. They may contain small amounts of free fatty acids and glycerol.
Description	The product varies from a pale yellow to pale brown oily liquid to a white or slightly off-white hard waxy solid. The solids may be in the form of flakes, powders or small beads.
Mono- and di-ester content	Not less than 70 %.
Free fatty acids	Not more than 3 % estimated as oleic acid.
Free glycerol	
	Not more than 7 %.
Total glycerol	Not more than 7 %. Not less than 16 % and not more than 33 %.
Total glycerol Polyglycerols	

Sulphated ash Not more than 0.5 % determined at 800 \pm 25 °C.

Note: These criteria are based on the product without E 470.

E 472 (a) — Acetic acid esters of mono- and diglycerides of fatty acids

Chemical description	Esters of glycerol with acetic acid and fatty acids occurring in
	food fats. They may contain small amounts of free glycerol, free
	fatty acids, free acetic acid and free glycerides.

Description	Clear, mobile liquids to solids, from white to pale yellow in colour.	
Total acetic acid content	Not less than 9 % and not more than 32 %.	
Free fatty acids (and acetic acid)	Not more than 3 % estimated as oleic acid.	
Free glycerol	Not more than 2 %.	
Total glycerol	Not less than 14 % and not more than 31 %.	
Sulphated ash	Not more than 0.5 % determined at 800 \pm 25 °C.	
E 472 (b) — Lactic acid esters	of mono- and diglycerides of fatty acids	
Chemical description	Esters of glycerol with lactic acid and fatty acids occurring in food fats. They may contain small amounts of free glycerol, free fatty acids, free lactic acid and free glycerides.	
Description	Soft to hard waxy solids.	
Total lactic acid content	Not less than 13 % and not more than 45 %.	
Free fatty acids	Not more than 3 % estimated as oleic acid.	
Free glycerol	Not more than 2 %.	
Total glycerol	Not less than 13 % and not more than 30 %.	
Sulphated ash	Not more than 0.5 % determined at 800 \pm 25 °C.	
Note: These criteria are based or	the product without E 470.	
E 472 (c) — Citric acid esters of	of mono- and diglycerides of food fatty acids	
Chemical description	Esters of glycerol with citric acid and fatty acids occurring in food oils and fats. They may contain small amounts of free glycerol, free fatty acids, free citric acid and free glycerides. They may be partially or wholly neutralized with sodium hydroxide or with potassium hydroxide.	
Description	Yellowish or light brown liquids to waxy solids or semi-solids.	
Total citric acid content	Not less than 13 % and not more than 50 %.	
Free fatty acids	Not more than 3 % estimated as oleic acid.	
Free glycerol	Not more than 2 %.	
Total glycerol	Not less than 11 % and not more than 29 %.	
Sulphated ash	Not more than 0.5 % for the non-neutralized products and not more than 10.0 % for the partially or wholly neutralized products determined at 800 \pm 25 °C.	
pH of a 1 % solution	Not less than 3 and not more than 7.3.	
E 472 (d) — Tartaric acid esters of mono- and diglycerides of food fatty acids		
Chemical description	Esters of glycerol with tartaric acid (E 334) and fatty acids occurring in food fats. They may contain small amounts of free glycerol, free fatty acids, free tartaric acid and free glycerides.	
Description	Sticky viscous yellowish liquids to hard yellow waxes.	
Total tartaric acid content	Not less than 15 % and not more than 50 %.	
Free fatty acids	Not more than 3 % estimated as oleic acid.	
Free glycerol	Not more than 2 %.	
Total glycerol	Not less than 12 % and not more than 29 %.	
Sulphated ash	Not more than 0.5 % determined at 800 \pm 25 °C.	

E 472 (e) — Mono- and diacetyl tartaric acid esters of mono- and diglycerides of fatty acids

▼B		
_	Chemical description	Esters of glycerol with mono- and diacetyl tartaric acids (obtained from E 334 tartaric acid) and fatty acids occurring in food fats. They may contain small amounts of free glycerol, free fatty acids, free tartaric and acetic acids and their combinations, and free glycerides.
	Description	Sticky viscous liquids through a fat-like consistency to yellow waxes which hydrolyse in moist air to liberate acetic acid.
	Total tartaric acid content	Not less than 10 % and not more than 40 %.
	Total acetic acid content	Not less than 8 % and not more than 32 %.
	Free fatty acids	Not more than 3 % estimated as oleic acid.
	Free glycerol	Not more than 2 %.
	Total glycerol	Not less than 11 % and not more than 28 %.
	Sulphated ash	Not more than 0.5 % determined at 800 \pm 25 °C.
	E 472 (f) — Mixed acetic and ta	rtaric acid esters of mono- and diglycerides of fatty acids
	Chemical description	Esters of glycerol with acetic and tartaric (E 334) acids and fatty acids occurring in food fats. They may contain small amounts of free glycerol, free fatty acids, free acetic and tartaric acids, and free glycerides.
	Description	Clear mobile liquids to solids, from white to pale yellow in colour.
	Total acetic acid	Not less than 10 % and not more than 20 %.
	Total tartaric acid	Not less than 20 % and not more than 40 %.
	Free acetic acid	Not less than 5.5 $\%$ and not more than 8.5 %.
	Free tartaric acid	Not more than 1 %.
	Free fatty acids	Not more than 3 % estimated as oleic acid.
	Free glycerol	Not more than 2 %.
	Total glycerol	Not less than 12 % and not more than 27 %.
	Sulphated ash	Not more than 0.5 % determined at 800 \pm 25 °C.
	E — 473 Sucrose esters of fatty	acids
	Chemical description	Essentially the mono- and di-esters of sucrose with fatty acids occurring in food fats. They may be prepared from sucrose and the methyl and ethyl esters of food fatty acids or by extraction from sucroglycerides. \blacktriangleright M3 No organic solvent other than dimethylsulphoxide, dimethylformamide, ethyl acetate, isopropanol, isobutanol and methylethylketon may be used in their preparation.
	Description	Soft solids, stiff gels or white to greyish-white powders.
	Total sucrose fatty acid ester content	Not less than 80 %.
	Total glyceride content	Not more than 20 %.
	Free sucrose content	Not more than 5 %.
	Free fatty acid content	Not more than 3 % estimated as oleic acid.
	Sulphated ash	Not more than 2 % determined at 800 \pm 25 °C.
▼ <u>M2</u>	Dimethylsulphoxide content	Not more than 2 mg/kg.
▼ <u>B</u>	Dimethylformamide content	Not more than 1 mg/kg.

▼ <u>B</u>	Methanol content	Not more than 10 mg/kg.
▼ <u>M2</u>	Isobutanol content	Not more than 10 mg/kg.
▼ <u>M3</u>	Methylethylketon content	Not more than 10 mg/kg.
	Total ethyl acetate and isopro- panol content	Not more than 350 mg/kg singly or in combination.
	Note: These criteria are based on the product without E 470.	
	E 474 — Sucroglycerides	
	Chemical description	Sucroglycerides are produced by reacting sucrose with an edible fat or oil to produce a mixture of essentially mono- and di-esters of sucrose and fatty acids together with residual mono-, di- and tri-glycerides from that fat or oil. \blacktriangleright <u>M1</u> No organic solvents shall be used in their preparation other than cyclohexane, dimethylformamide, ethyl acetate, isobutano land isopropanol.
	Description	Soft solid masses, stiff gels or white to off-white powders.
	Total sucrose fatty acid ester content	Not less than 40 $\%$ and not more than 60 $\%$.
	Total glyceride content	Not less than 40 $\%$ and not more than 60 %.
	Free sucrose content	Not more than 5 %.
	Free fatty acid content	Not more than 3 % estimated as oleic acid.
	Sulphated ash	Not more than 2 % determined at 800 \pm 25 °C.
	Dimethylformamide content	Not more than 1 mg/kg.
	Methanol content	Not more than 10 mg/kg.
	Total ethyl acetate and isopro- panol content	Not more than 350 mg/kg singly or in combination.

▼M1

Total cyclohexane and isobutanol Not more than 10 mg/kg singly or in combination. content

▼<u>B</u>

Note: These criteria are based on the product without E 470.

E 475 — Polyglycerol esters of non-polymerized fatty acids

Chemical description	Polyglycerol esters of fatty acids are produced by the esterifica- tion of polyglycerol with food fats or with fatty acids occurring in food fats. The polyglycerol moiety is predominantly di-, tri- and tetra-glycerol and contains not more than 10 % of polygly- cerols equal to or higher than heptaglycerol.
Description	Yellow or light brown liquids or semi-solids.
Total fatty acid ester content	Not less than 90 %.
Free fatty acids	Not more than 6 % estimated as oleic acid.
Total glycerol and polyglycerol	Not less than 18 % and not more than 60 %.
Free glycerol and polyglycerol	Not more than 7 %.
Sulphated ash	Not more than 0.5 % determined at 800 \pm 25 °C.

Note: These criteria are based on the product without E 470.

E 477 — Propane-1,2-diol esters of fatty acids

Chemical description Consists chiefly of mixtures of propane-1,2-diol mono- and di-esters of fatty acids occurring in food fats. The alcohol moiety is exclusively propane-1,2-diol together with dimer and traces of trimer. Organic acids other than food fatty acids are absent.

Description	Waxy white flakes, beads or solids.
Total fatty acid ester content	Not less than 85 %.
Free propane-1,2-diol	Not more than 5 %.
Dimer and trimer of propane-1,2- diol	▶ <u>M1</u> Not more than 0.5 %. \triangleleft
Free fatty acids	Not more than 6 % estimated as oleic acid.
Sulphated ash	Not more than 0.5 % determined at 800 \pm 25 °C.
Total propane-1,2	Not less than 11 % and not more than 31 %.

Note: These criteria are based on the product without E 470.

E 481 — Sodium stearoyl-2-lactylate

Chemical description	A mixture of the sodium salts of stearoyl lactylic acids and minor amounts of sodium salts of other related acids, manufac- tured by the reaction of stearic acid and lactic acid. Other food fatty acids may also be present, free or esterified, due to their presence in the stearic acid used.	
Description	Cream coloured powder or brittle solid with a characteristic odour.	
Sodium content	Not less than 2.5 $\%$ and not more than 5 $\%.$	
Ester value	Not less than 90 and not more than 190 mg KOH/g.	
Total lactic acid (free and combined)	Not less than 15 $\%$ and not more than 40 $\%.$	
Acid value	Not less than 60 and not more than 130 mg KOH/g.	

E 482 — Calcium stearoyl-2-lactylate

Chemical description	A mixture of calcium salts of stearoyl lactylic acids with minor amounts of calcium salts of other related acids, manufactured by the reaction of stearic acid and lactic acid. Other food fatty acids may also be present, free or esterified due to their presence in the stearic acid used.	
Description	White or slightly yellowish powder or brittle solid with a characteristic odour.	
Calcium content	Not less than 1.0 $\%$ and not more than 5.2 %.	
Ester value	Not less than 125 and not more than 190 mg KOH/g.	
Total lactic acid (free and combined)	Not less than 15 $\%$ and not more than 40 %.	
Acid value	Not less than 50 and not more than 130 mg KOH/g.	
E 483 — Stearyl tartrate		
Chemical description	Stearyl tartrate is produced by the esterification of tartaric acid (E 334) with stearyl alcohol. It consists chiefly of the di-ester with minor amounts of mono-ester, tartaric acid and stearyl alcohol. Other esters may also be present due to the presence in the stearyl alcohol used of alcohols derived from food fatty acids other than stearic acid.	
Description	Cream coloured unctuous solid (at 25 °C).	
Total ester content	Not less than 90 %.	
Total tartaric acid content	Not less than 18 % and not more than 35 %.	
Unsaponifiable matter	Not less than 77 % and not more than 83 %.	
Unsaponifiable matter Melting range	Not less than 77 % and not more than 83 %. 67 to 77 °C.	

Iodine value	Not more than 4 (Wijs).
Acid value	Not more than 6 mg KOH/g.
Sulphated ash	Not more than 0.5 % determined at 800 \pm 25 °C.

(1) A special method of analysis is required to determine this.